



Energy research Centre of the Netherlands

Oxidation of Ethylbenzene in a millimeter G-L Taylor flow reactor

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Background

Structured reactors and energy saving

- In the Netherlands, Chemical & Refining Industry uses 40% of the total primary energy consumption with partial oxidation processes as large energy consumers.
- Structured reactors with Taylor flow (TF) (Fig. 2) have improved heat and mass transfer, enabling high selectivity and conversion
- Energy savings is possible because of decreasing of energy consumption down stream

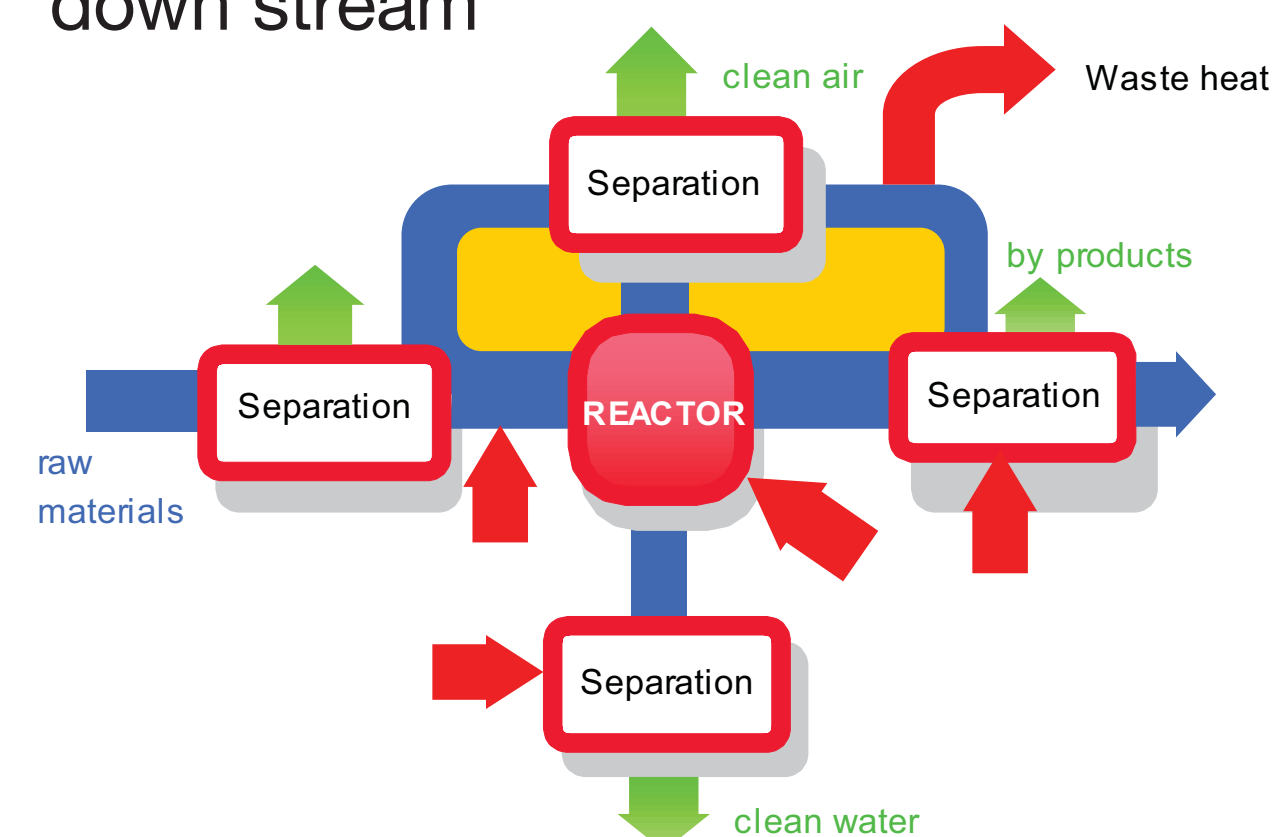


Figure 1: Energy usage in process industry

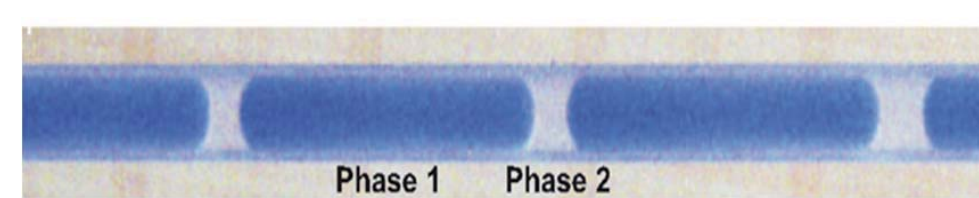


Figure 2: Taylor flow

Objective

- Demonstrating the strength of Taylor flow reactors in improving the selectivity in G-L reactions
- Develop and build a milli meter Taylor flow reactor equipment (mm TFR)
- Performing autocatalytic oxidation of ethylbenzene with pure ox

Structure of the project

- Conceptual design → safety analysis → P&ID → Building the mm TFR equipment
- Extended experimental programme on autocatalytic reaction.
 - ✓ Kinetic modelling to support building experimental matrix
 - ✓ Hydrodynamic experiments to define Taylor flow conditions
 - ✓ Performing oxidation experiments

Partial oxidation kinetics

Autocatalytic oxidation [2]

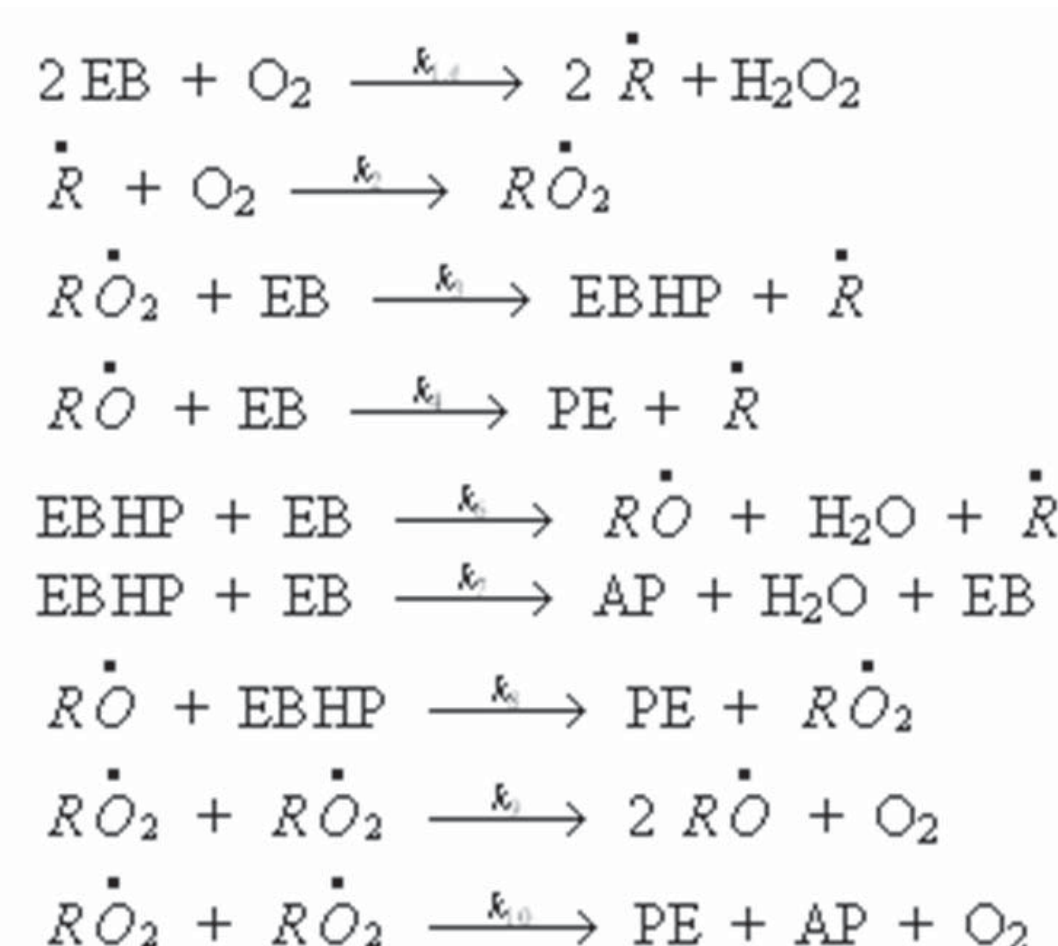


Figure 3: Model reaction for Ethylbenzene oxidation [1]

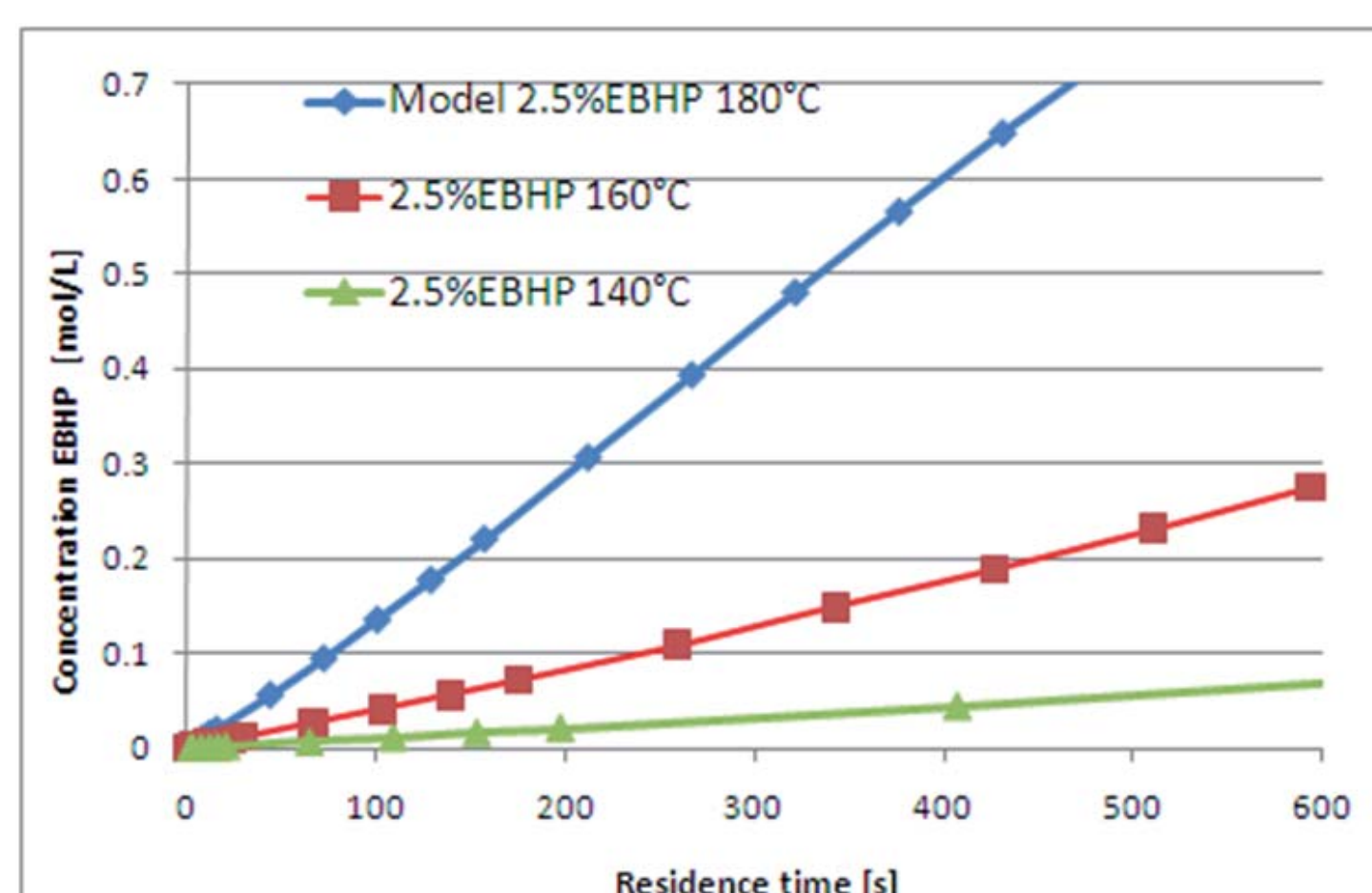


Figure 4: Operating window for temperature range 140 -180 °C

Flow Pattern map

Water-air (ambient)

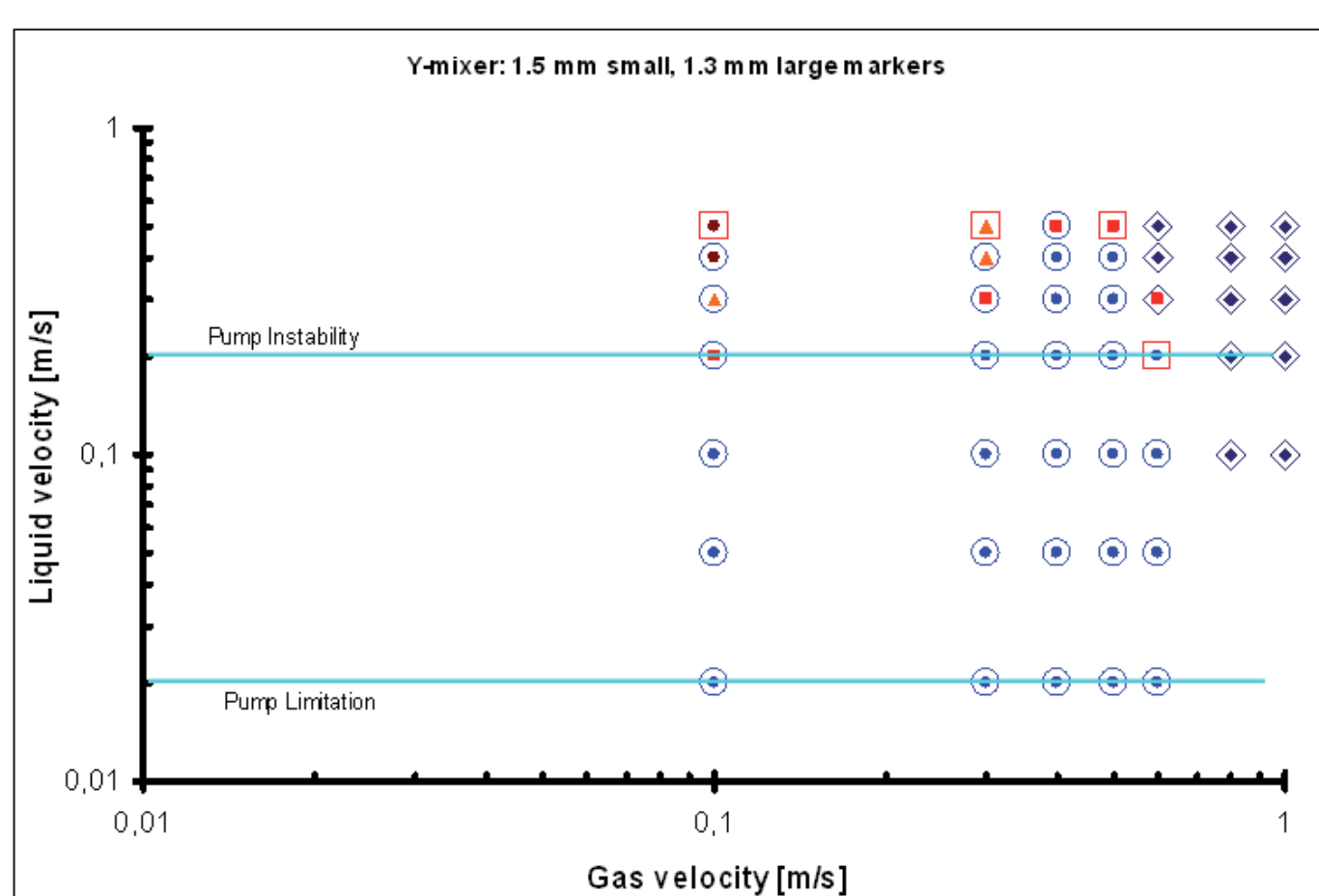


Figure 5: Operating window for fluid flow range: VG < 0.6 m/s, VL < 0.2 m/s

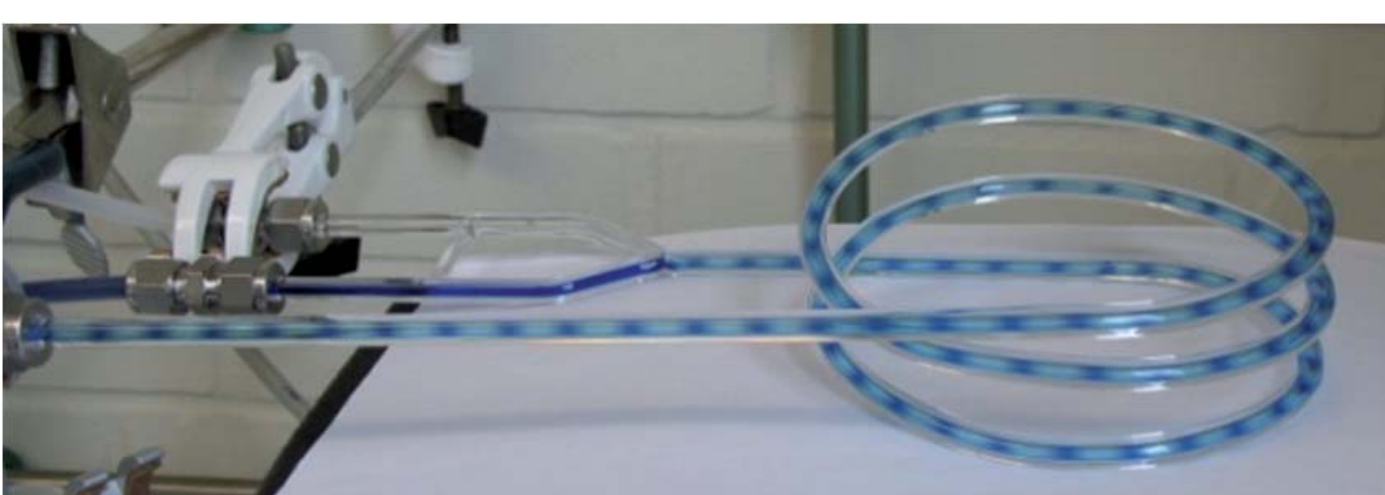


Figure 6: Hydrodynamic set-up

Results



Figure 7: Taylor flow reactor installation build at ECN

Scope
T = 300 C
P = 40 bar
F_L = 100 ml_r/min
F_G = 200 ml_r/min

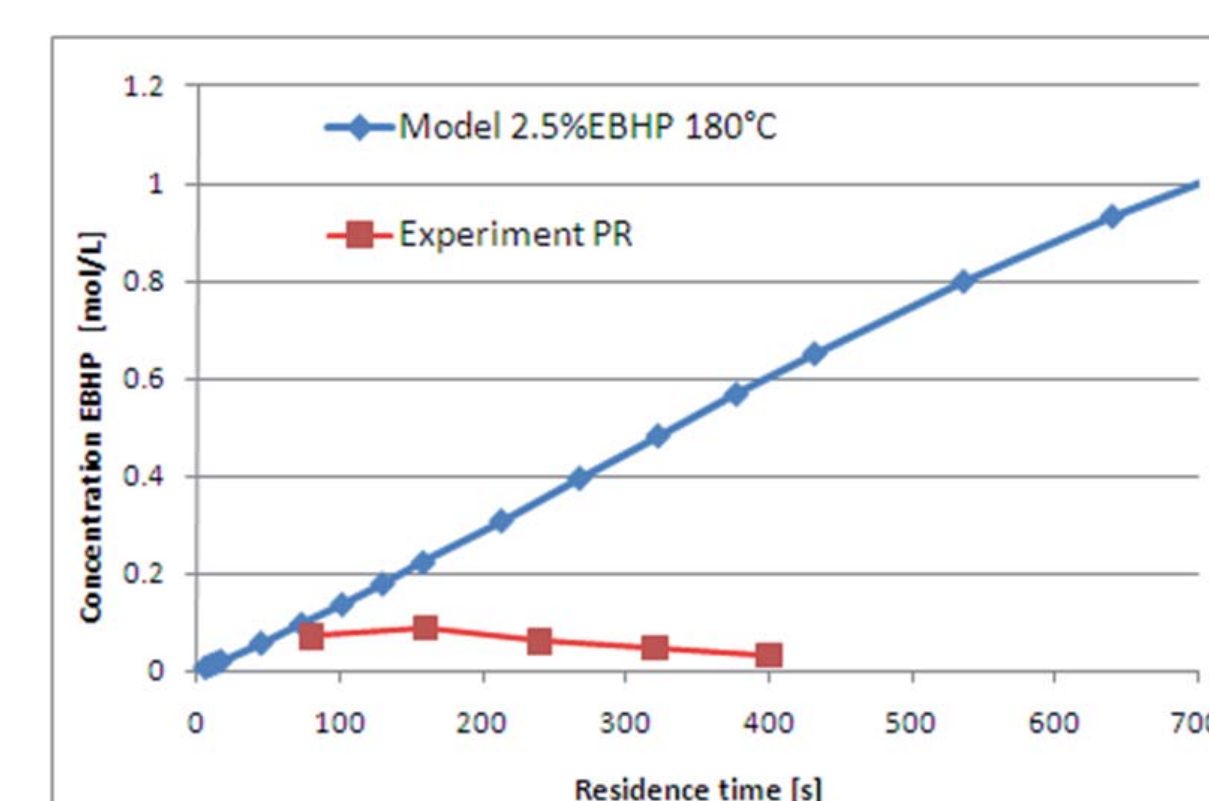
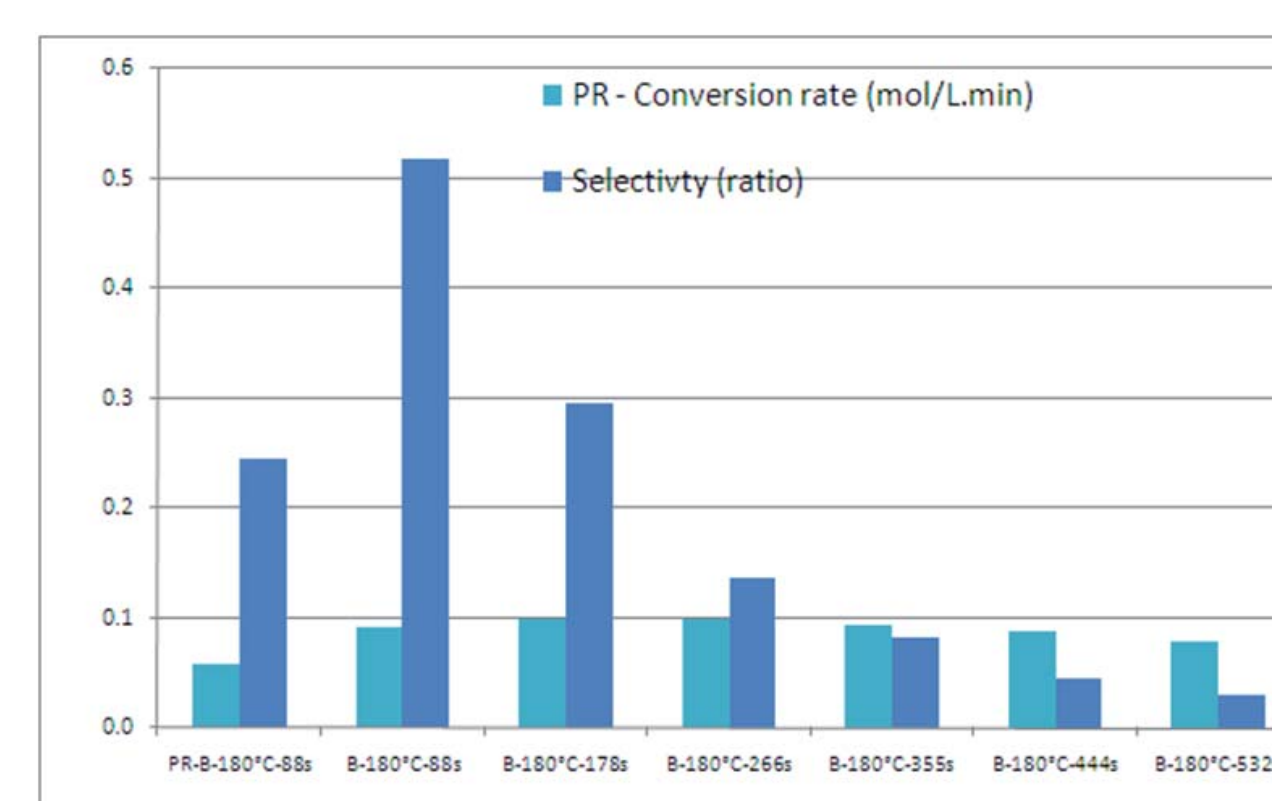


Figure 8: Conversion and selectivity at 180 °C compared with results of kinetic model

Conclusions

- Oxidation with pure oxygen can be performed safely in mm ECN TFR
- EBHP production is feasible and the results are in agreement with the model
- Prolonged processing leads to increased decomposition

Future work

Optimize process conditions and reactor specifications to eliminate increased decomposition

Acknowledgement

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