



Energy research Centre of the Netherlands

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ICG Institut
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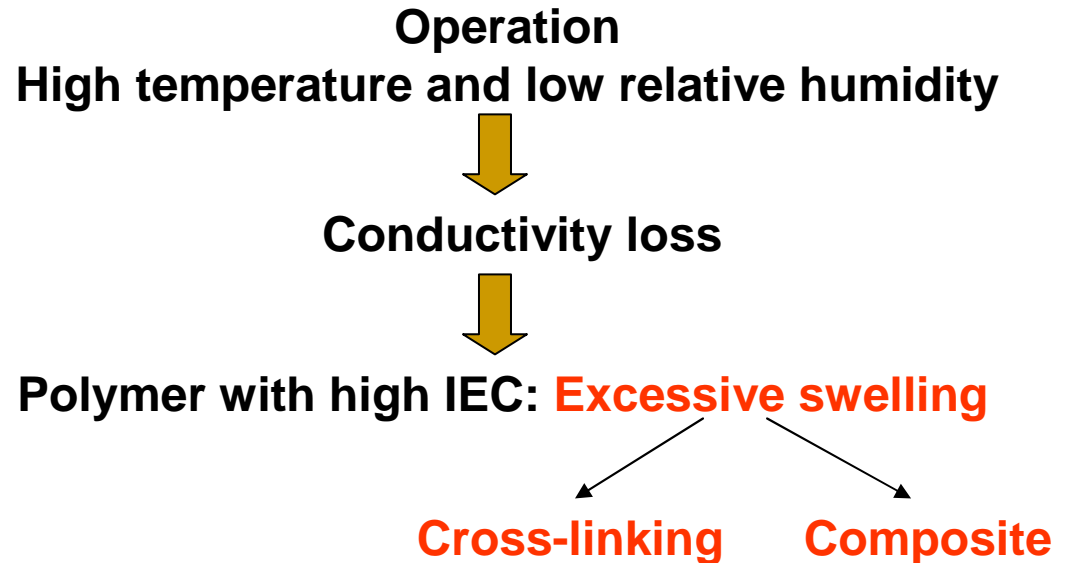


SIXTH FRAMEWORK PROGRAMME

Characteristics of Perfluorosulfonic Acid (PFSA) membranes

◆ Advantages

Excellent proton conductivity
High Chemical inertness
Good thermal stability
Good durability in fuel cell

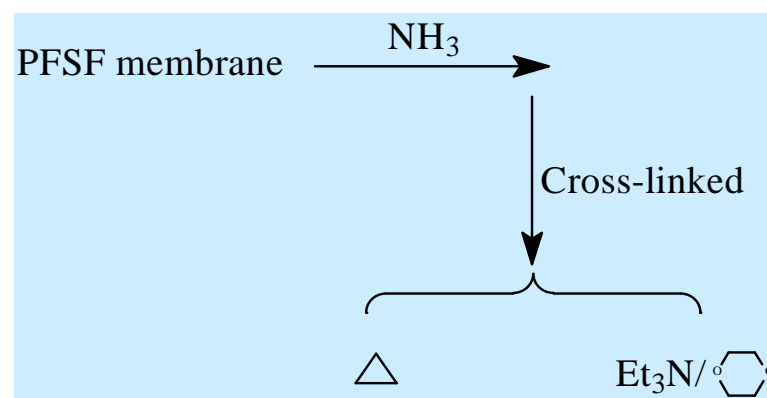
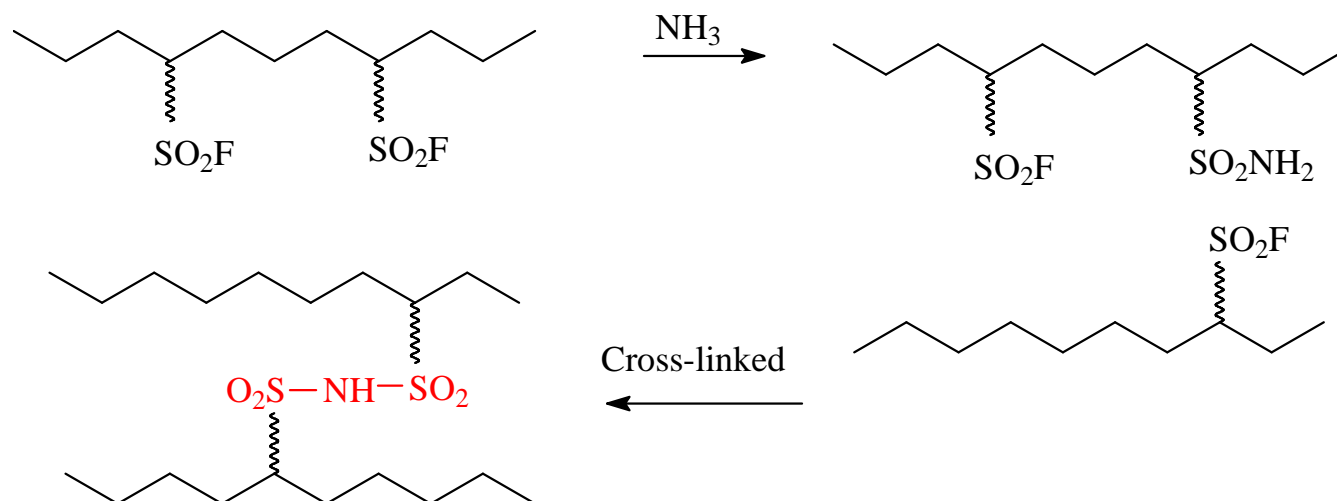


◆ Two methods for cross-linking PFSA membranes

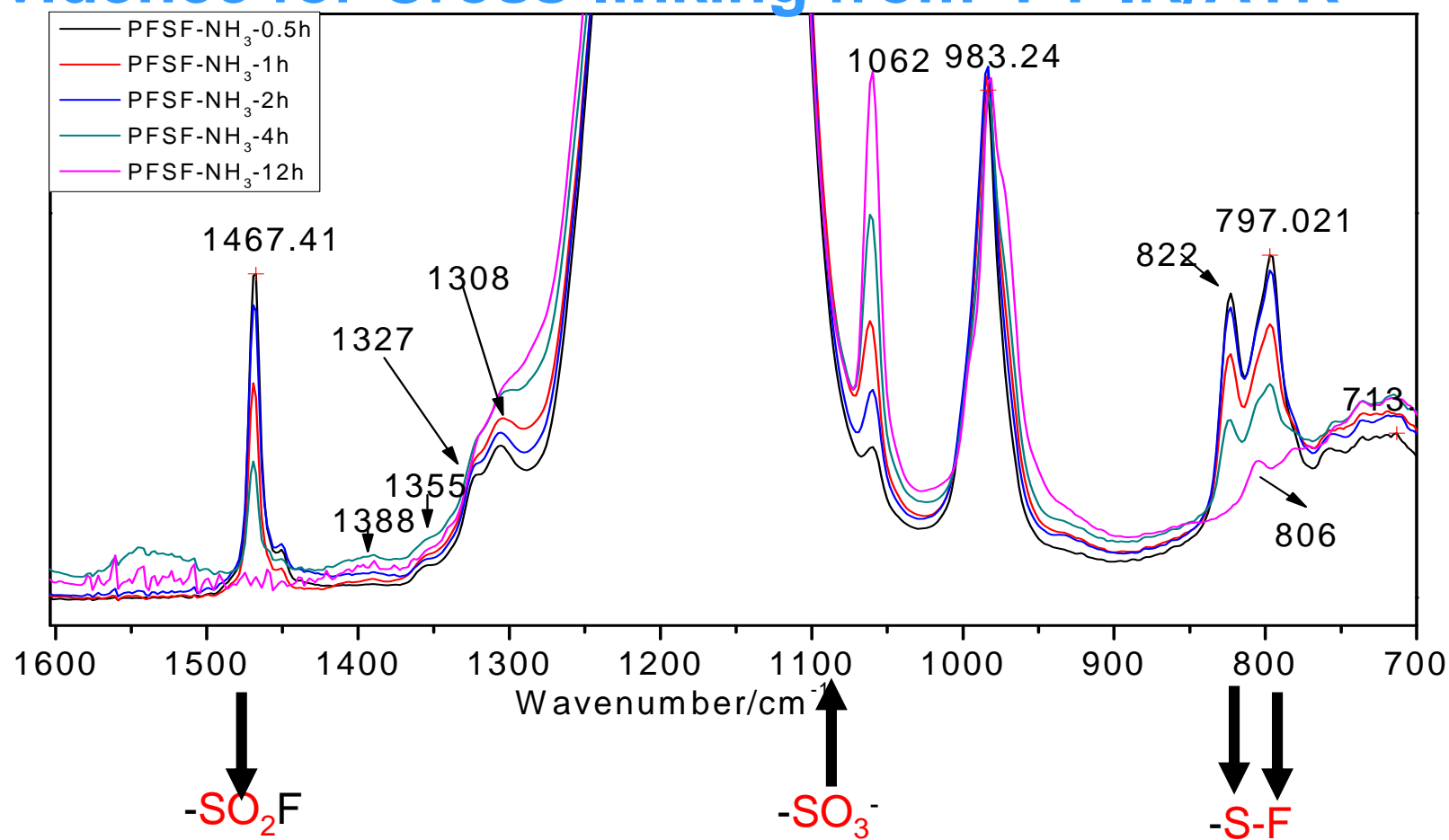
- Cross-linked PFSA membranes by formation of *sulfonimide*
- Cross-linked PFSA membranes by formation of *perfluoroalkyl*

◆ Inorganic component : *Zirconium phosphate (ZrP)*

◆ Cross-linked PFSA membranes by formation of Sulfonimide

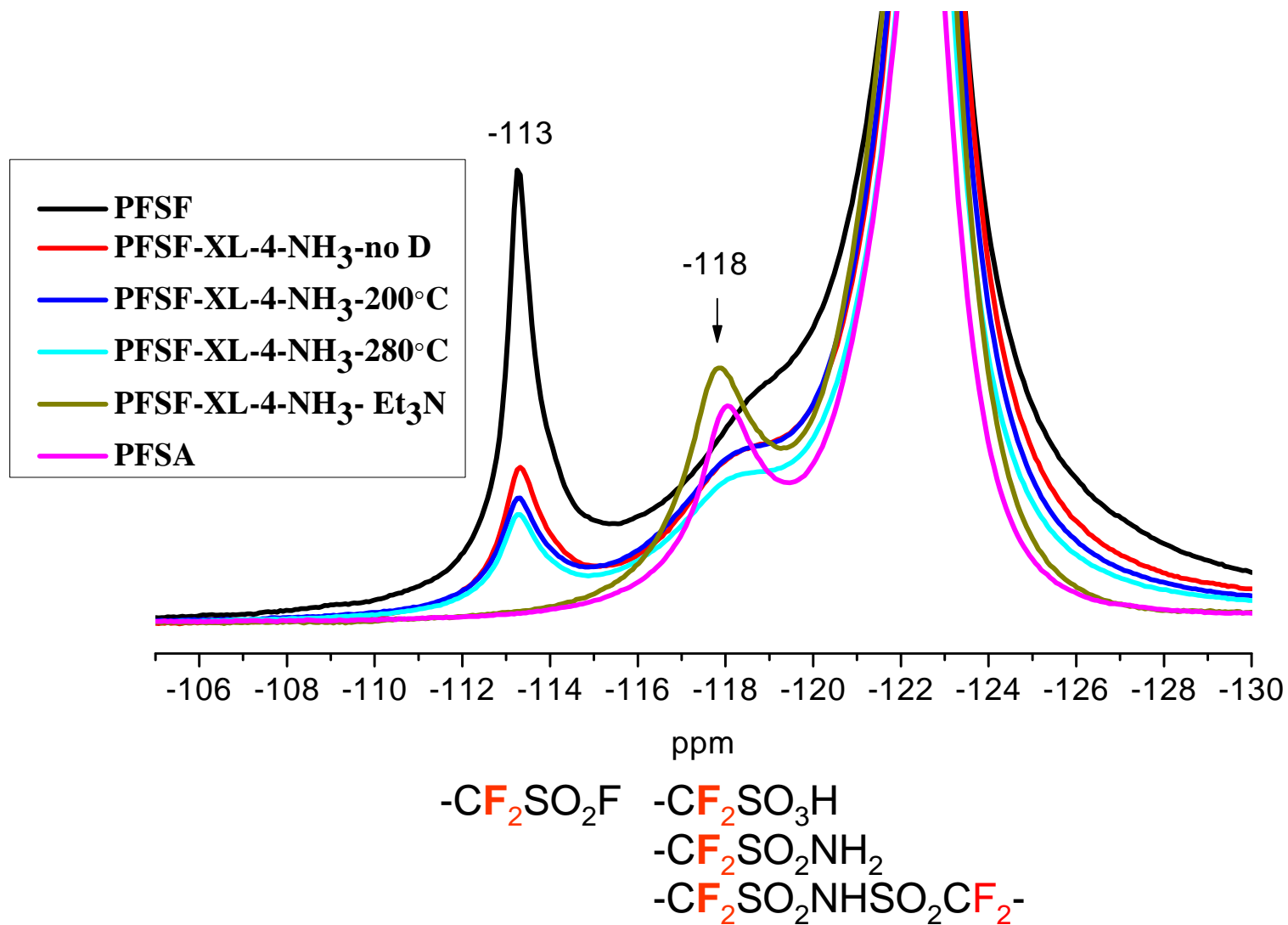


Evidence for Cross-linking from FT-IR/ATR

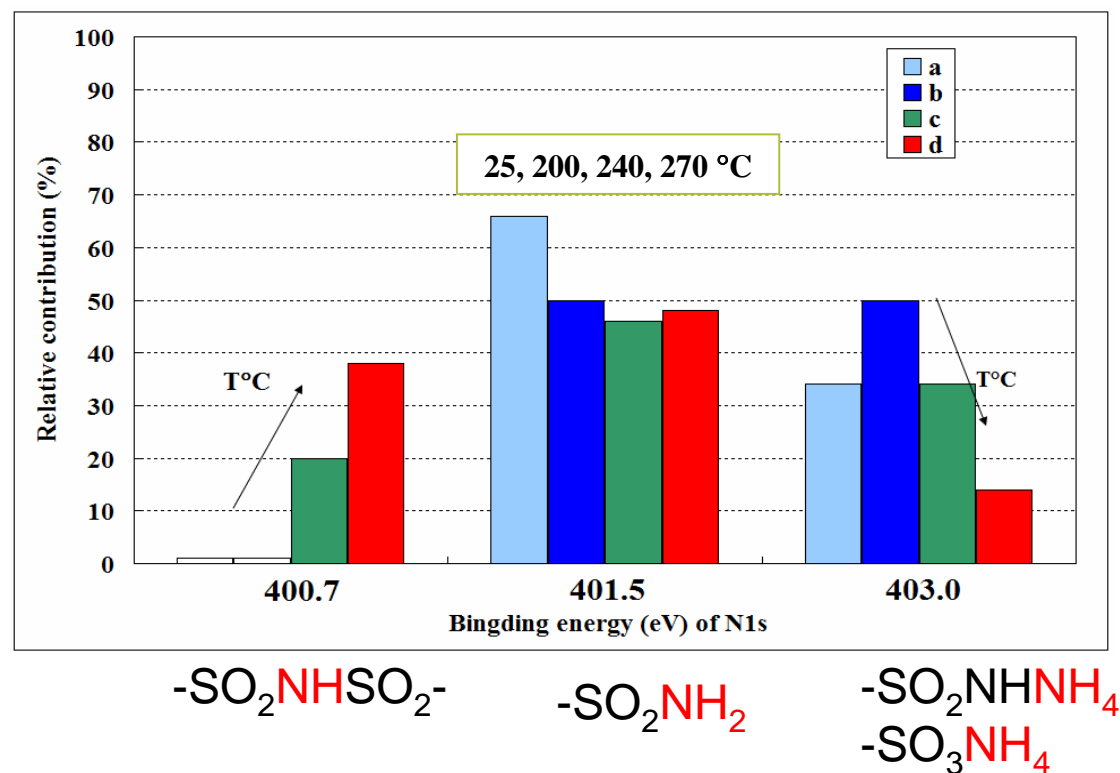
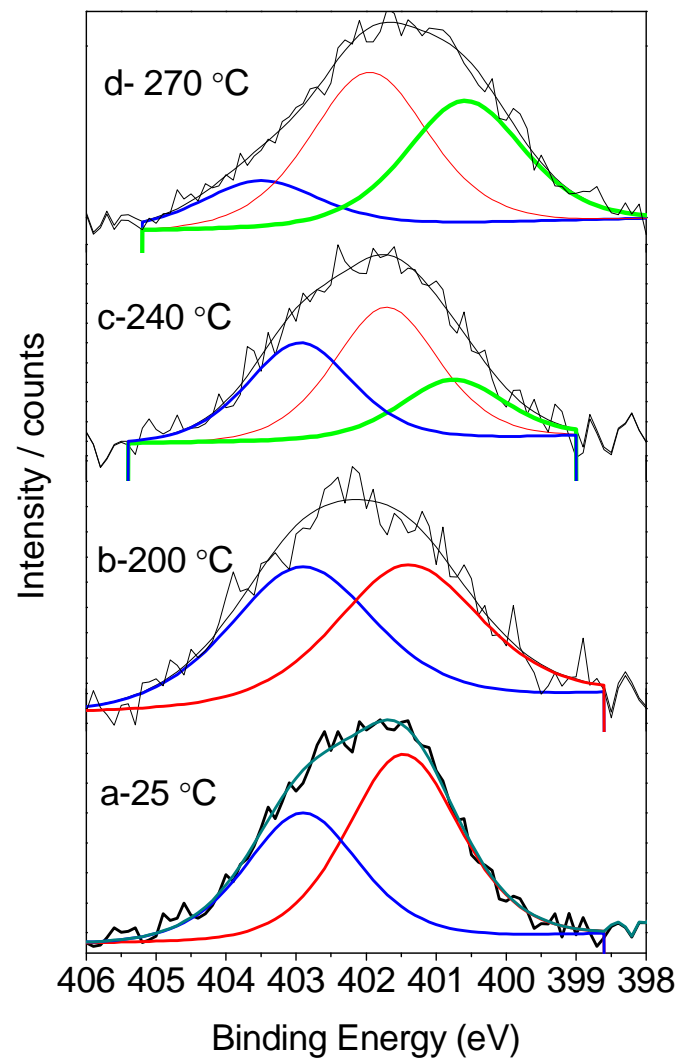


Reaction time ↗ -SO₂F ↘ 1467, 822 and 797 cm⁻¹
 ↗ -SO₃⁻ ↘ 1062 cm⁻¹

Evidence for Cross-linking from ^{19}F NMR

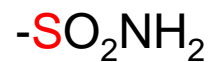
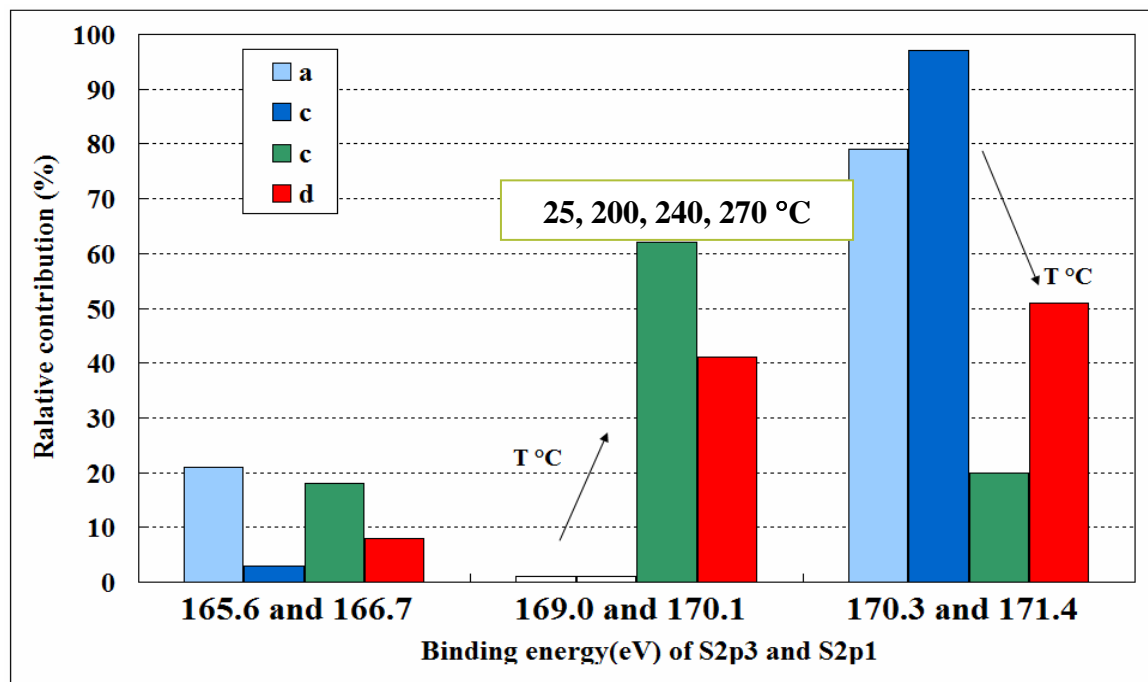
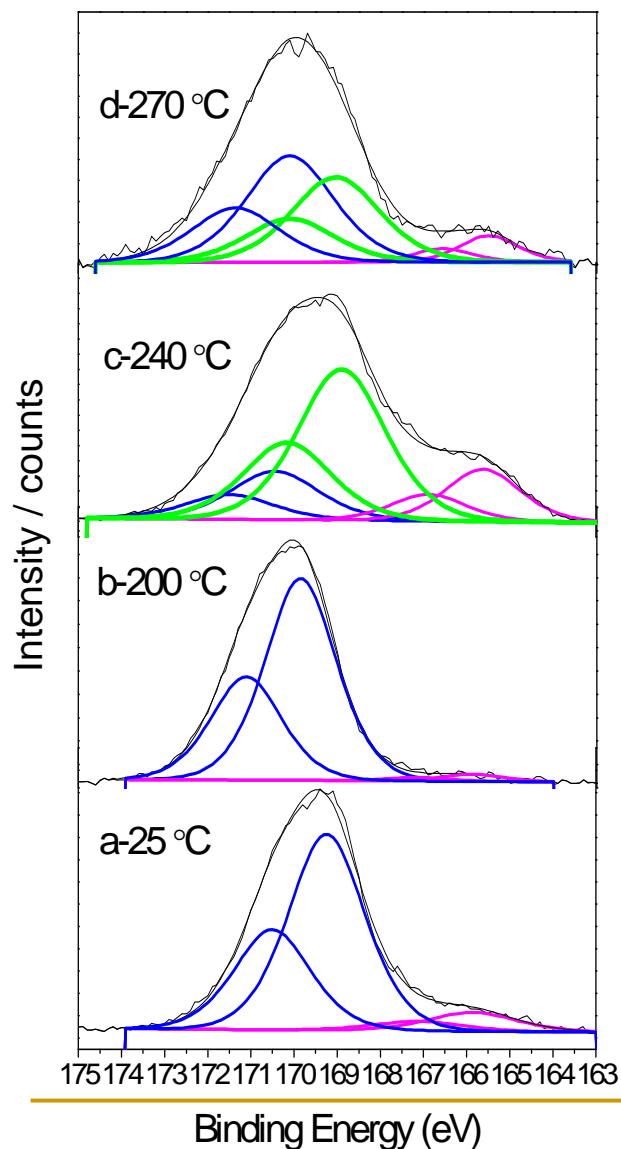


Evidence for Cross-linking from XPS- N 1s



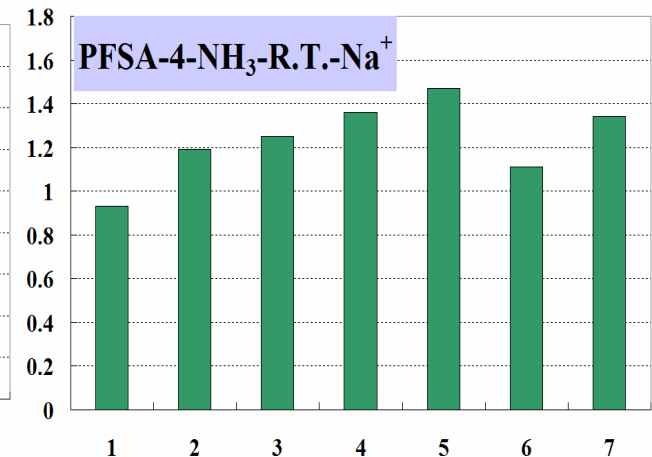
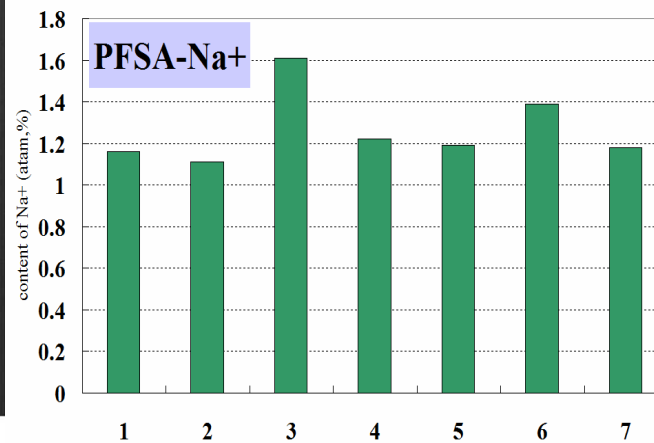
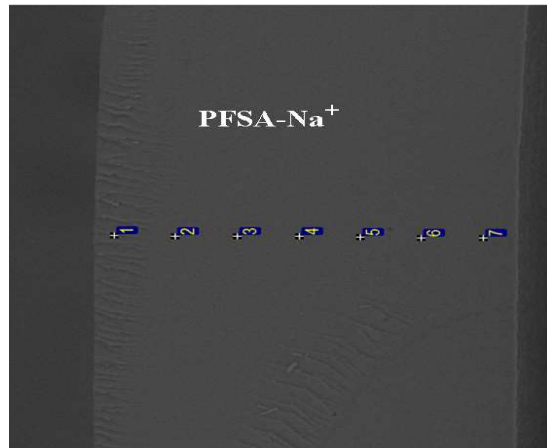
S. Caporali et al. Journal of Electron Spectroscopy and Related Phenomena, 2006, V.151, 4~8.

Evidence for Cross-linking from XPS- S 2p

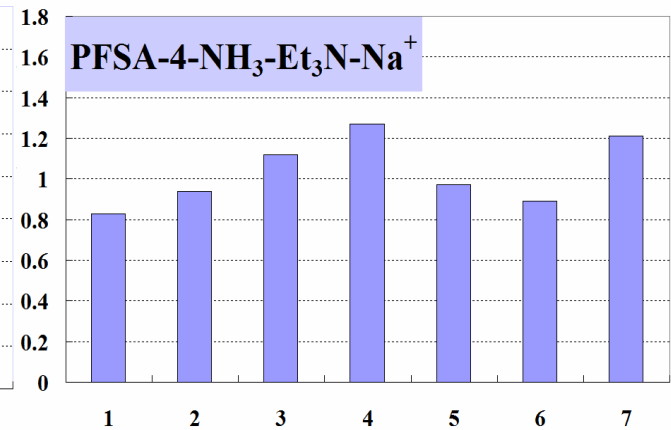
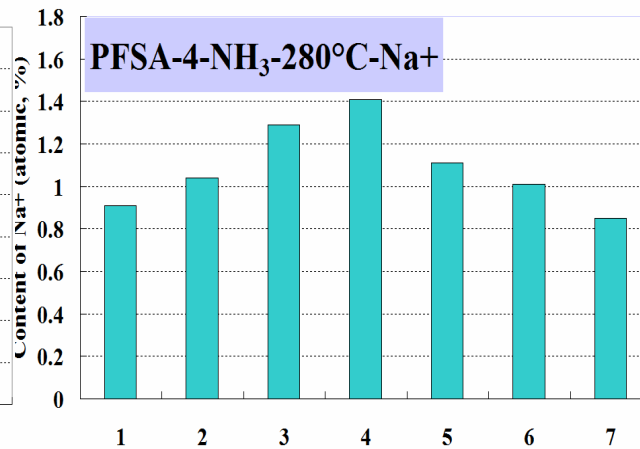
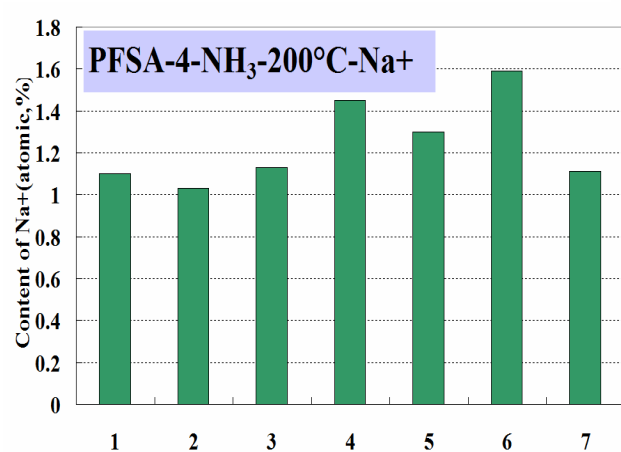


$2p_{1/2}$ $2p_{3/2}$

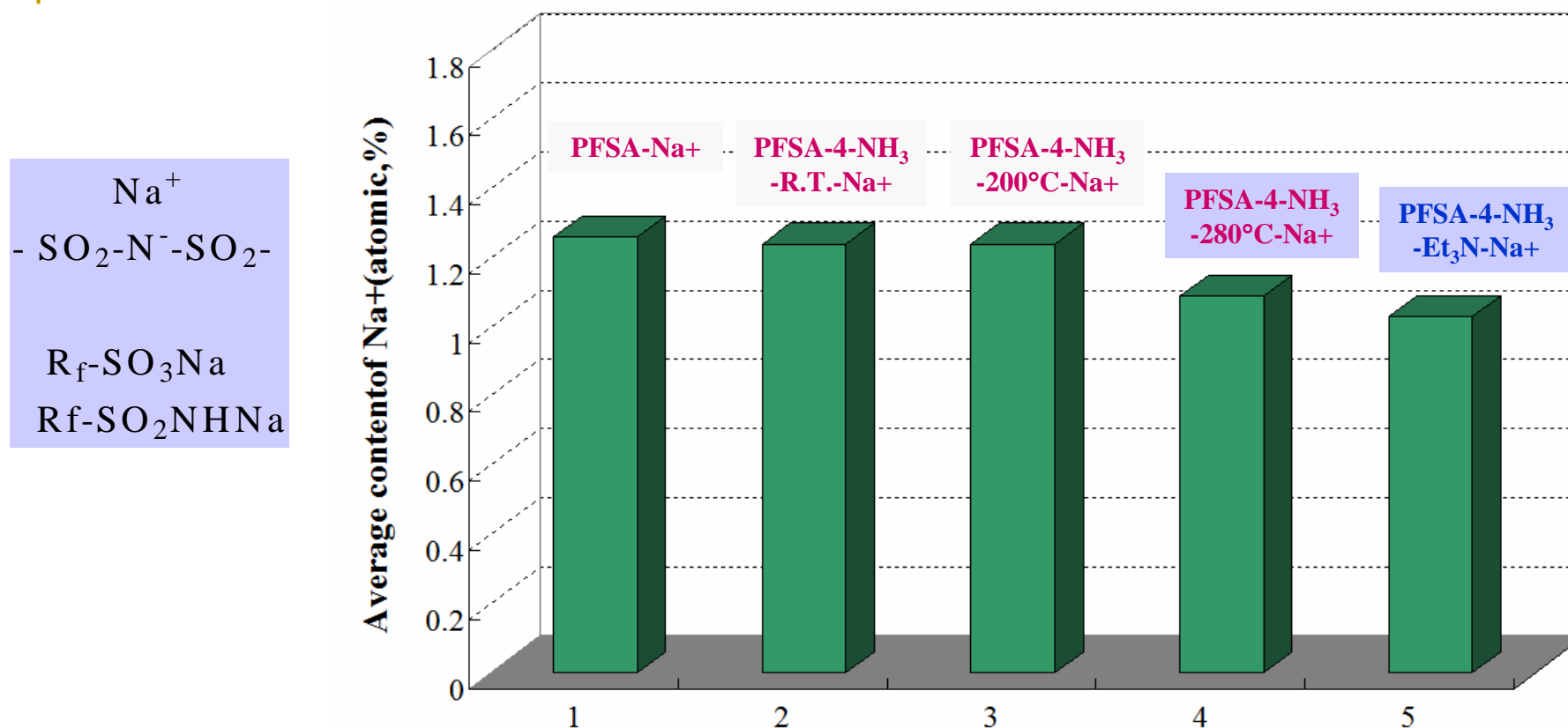
Evidence for Cross-linking - SEM-EDX



SEM micrograph of PFSA-Na⁺ and its sketch map of 7 point analysis across the membrane thickness.



Evidence for Cross-linking from SEM-EDX

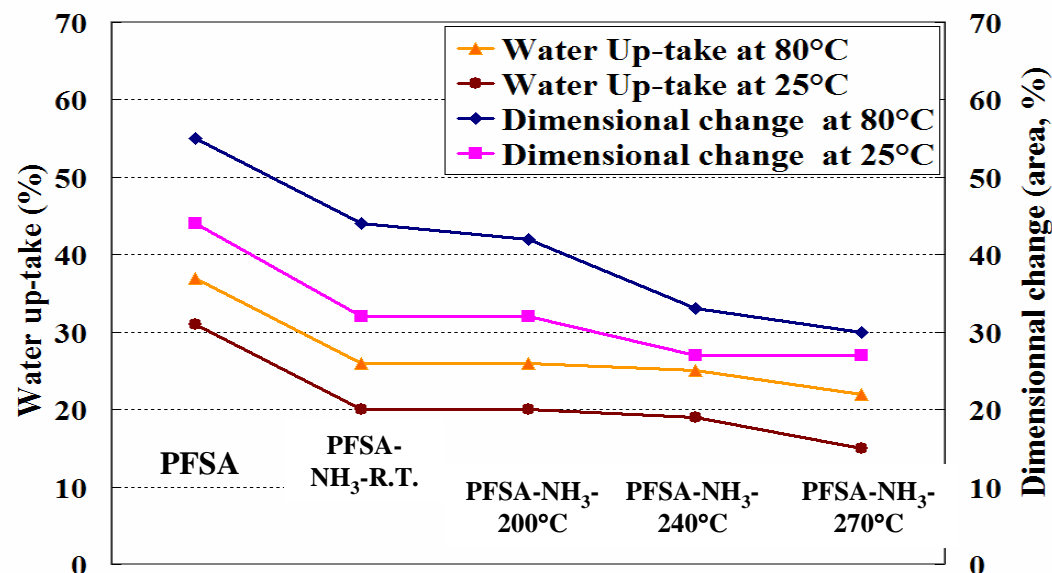


- Similar average Na⁺ content in Na-PFSA, Na-PFSA-NH₃-without Δ and Na-PFSA-NH₃-200
- Lower average Na⁺ content in membrane treated at 280 °C and Et₃N
- Gradient of Na⁺ content between surface (lower) and centre (higher) of membrane → higher degree of cross-linking at the membrane surfaces

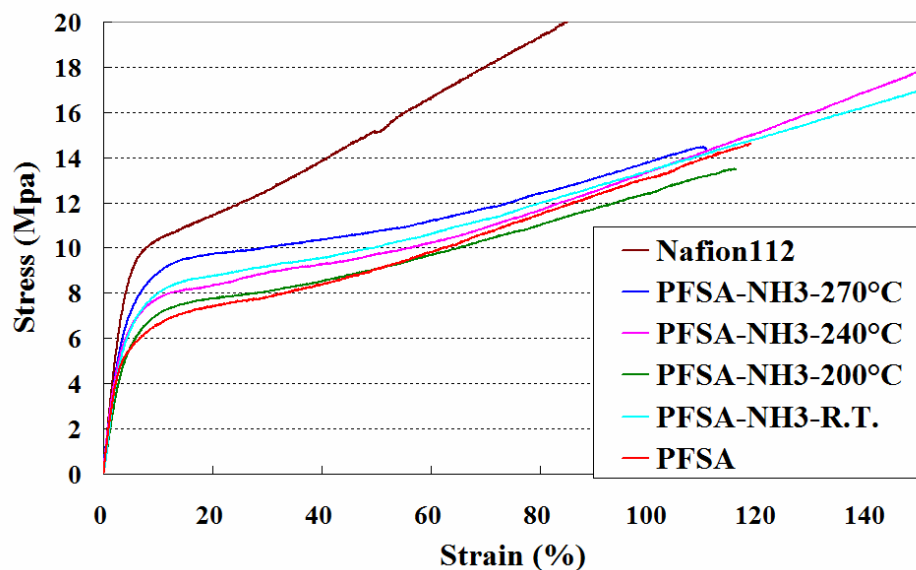
Properties of Cross-linked membranes

- *Water up-take and Dimensional change*
- *Mechanical Property*
- *Proton Conductivity*

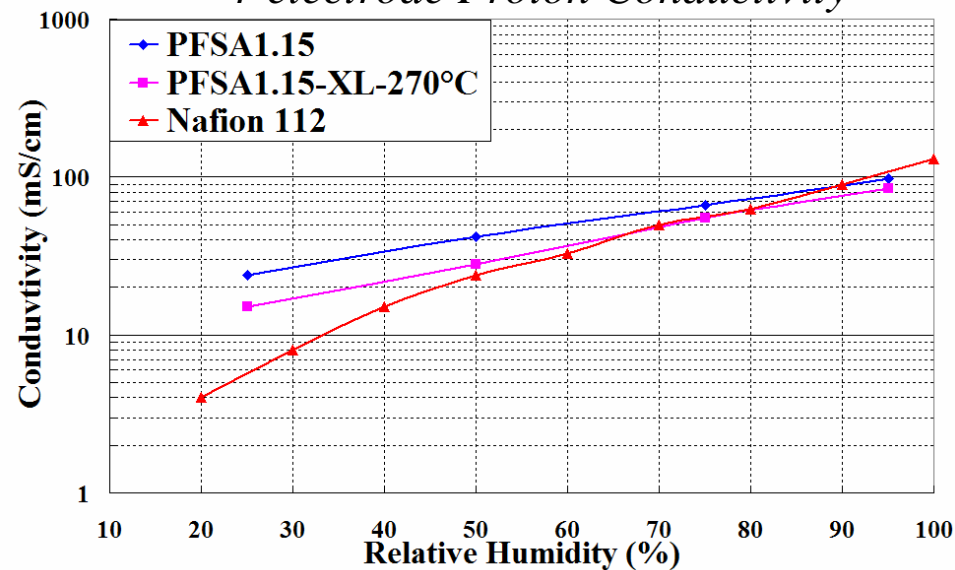
Water up-take and Dimensional change



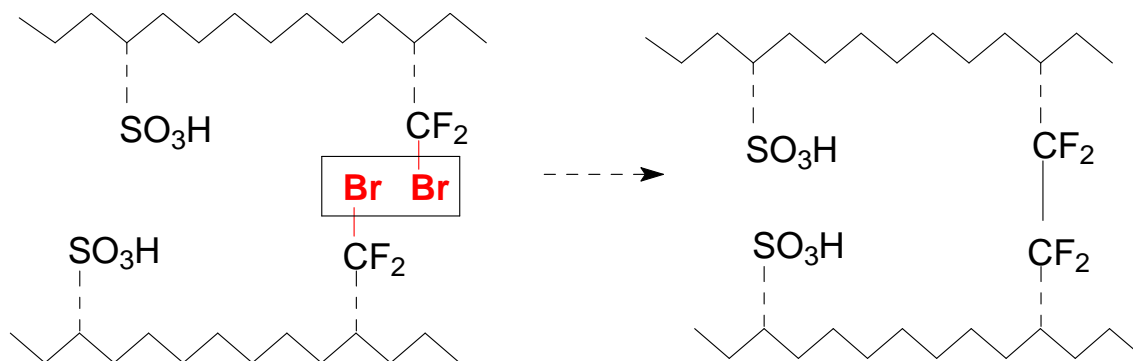
Mechanical Measurement



4-electrode Proton Conductivity



◆ Cross-linked PFSA membranes by formation of perfluoroalkyl

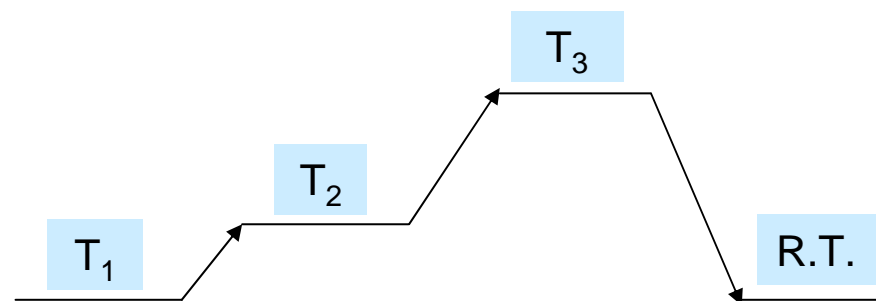


Cross-linking reaction: heat-induced Bromoalkyl

TGA of the resin

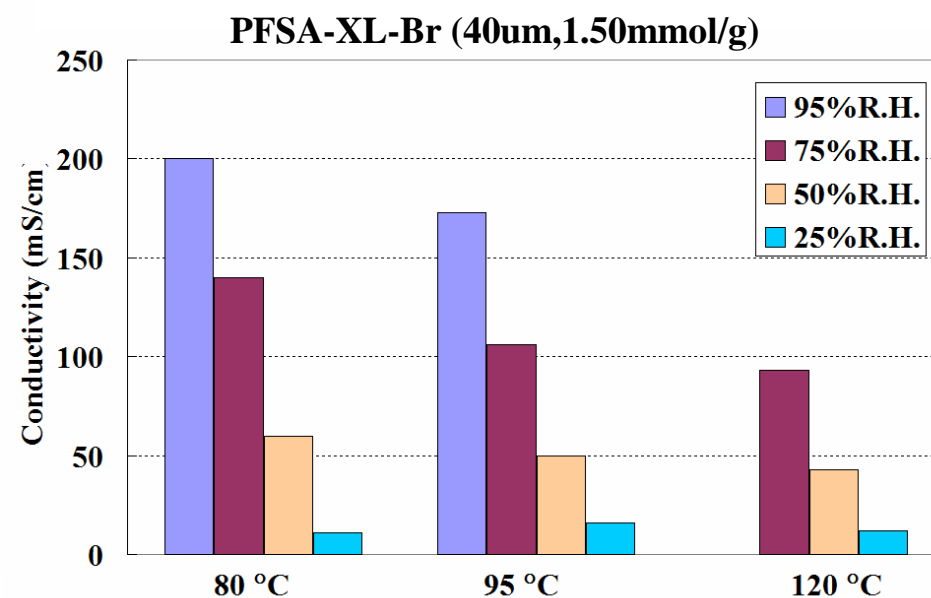
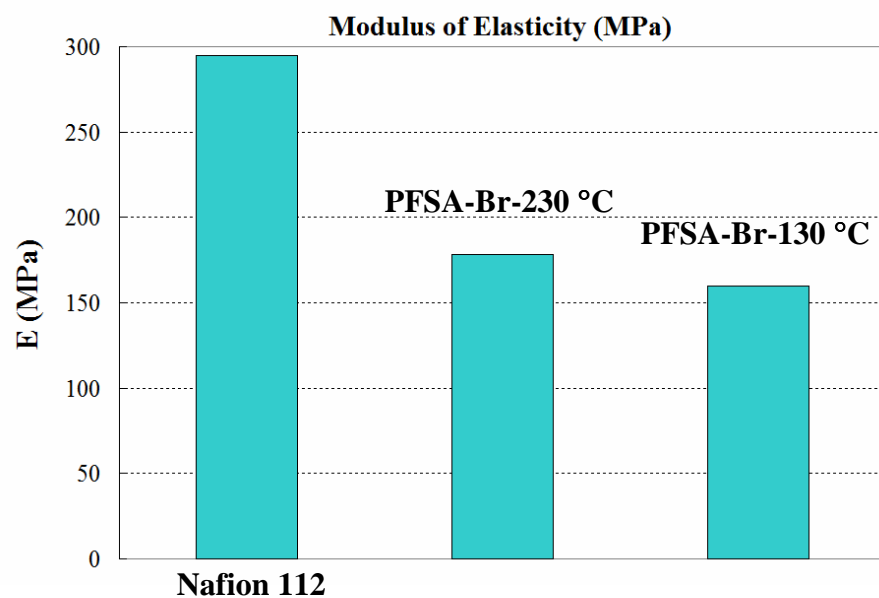
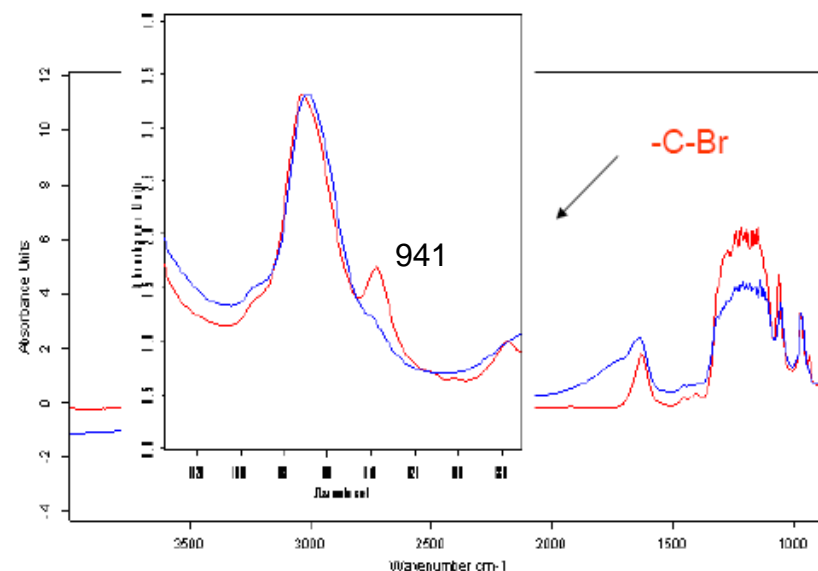
Temperature (°C)	Mass lost (w %)
20 ~ 100 (drying for 1h)	6.0 (water)
115 ~ 230	2.8 (Br₂+H₂O)
256 ~ 600	Decomposition

Process of thermal treatment by solution casting

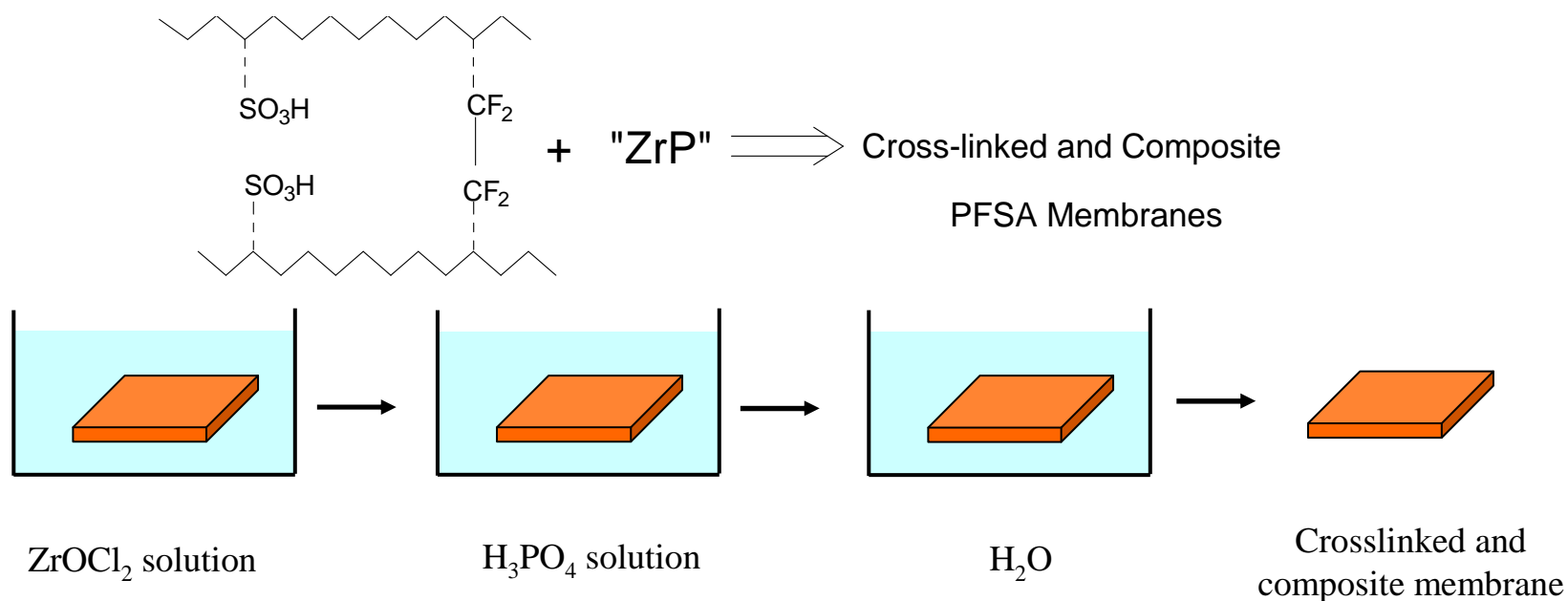


Characterisation of Cross-linked PFSA

- *IR*
- *Modulus of Elasticity*
- *Proton Conductivity*



Preparation of composite membrane by ion-exchange/precipitation

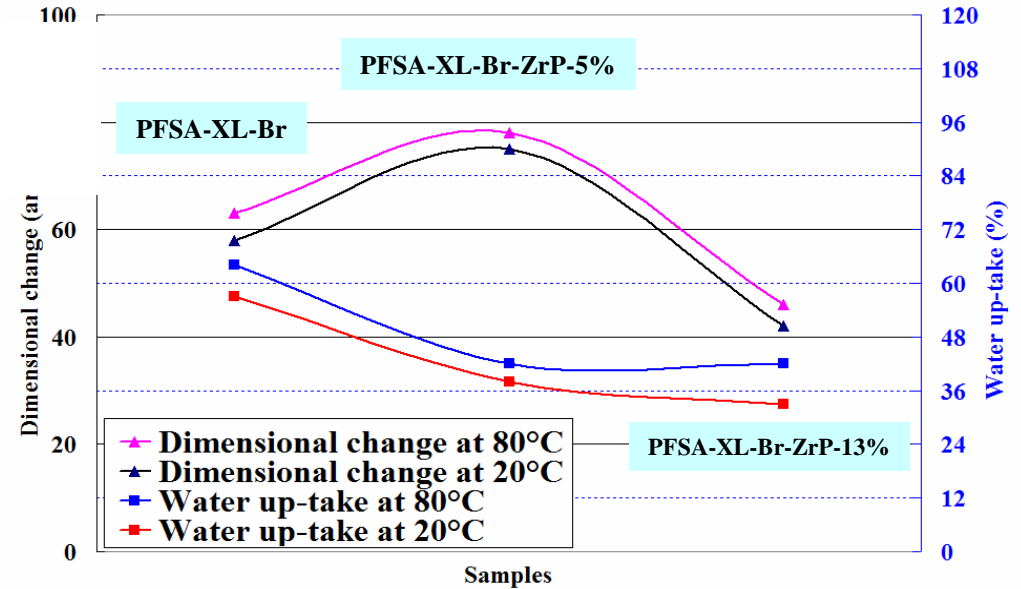


Samples	IEC (mmol/g)	Thickness (μm)	Content of "ZrP" (w%)
PFSA-XL-Br-ZrP-1	1.50	40	5
PFSA-XL-Br-ZrP-2	1.50	40	13

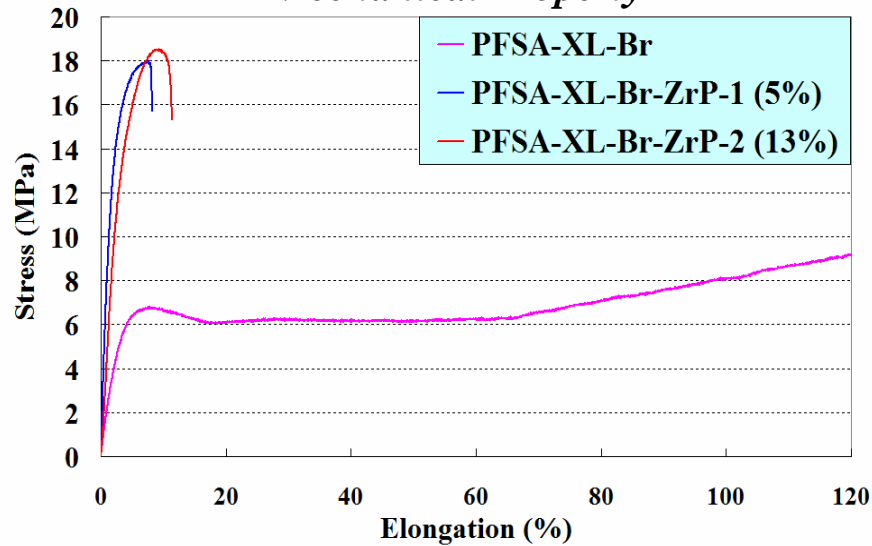
Properties of cross-linked and composite membranes

- *Water up-take and Dimensional change*
- *Mechanical Property*
- *Proton conductivity*

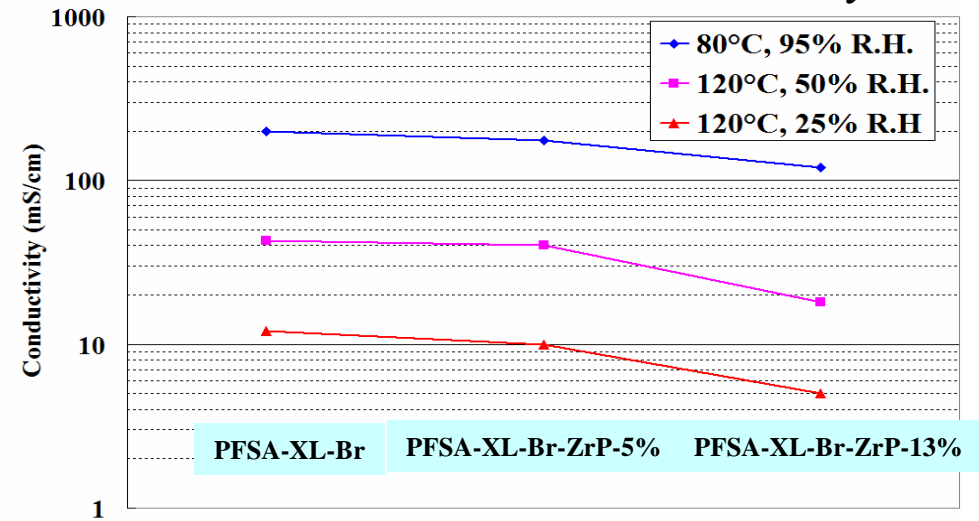
Dimensional change and Water up-take



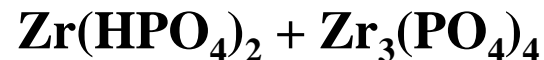
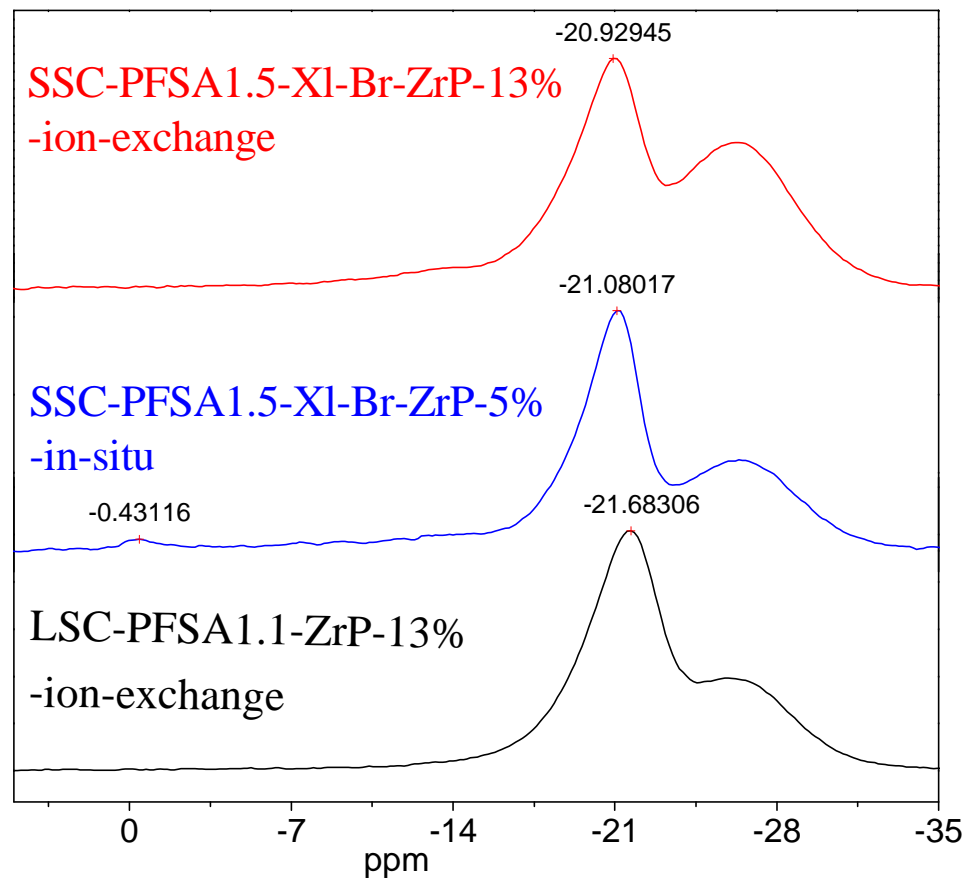
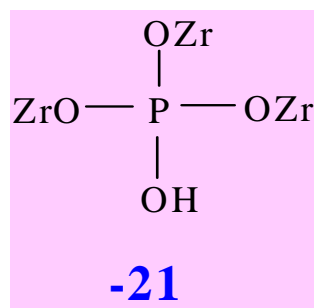
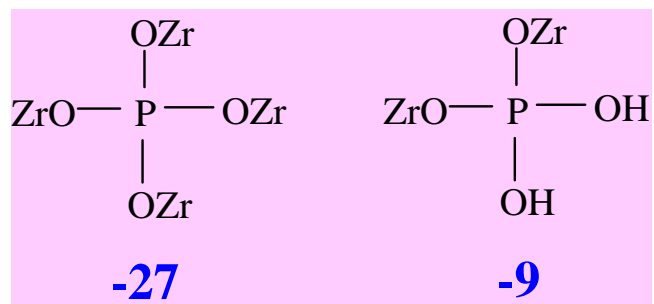
Mechanical Property



4 - Electrode Proton Conductivity



“ZrP” structure : ^{31}P NMR



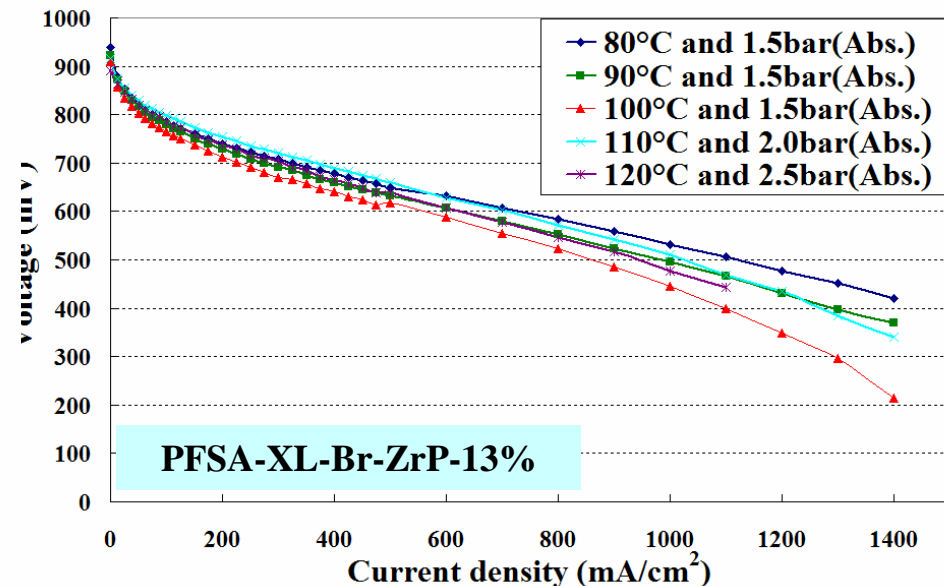
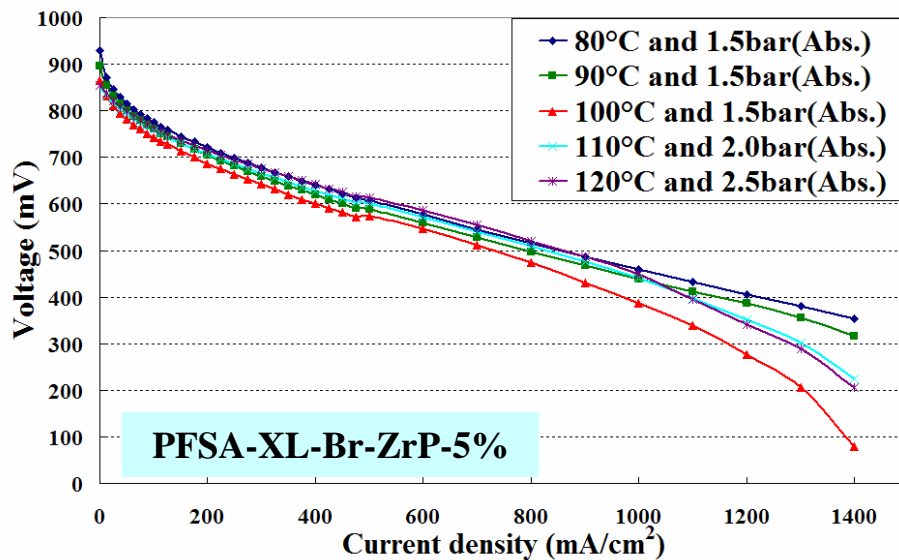
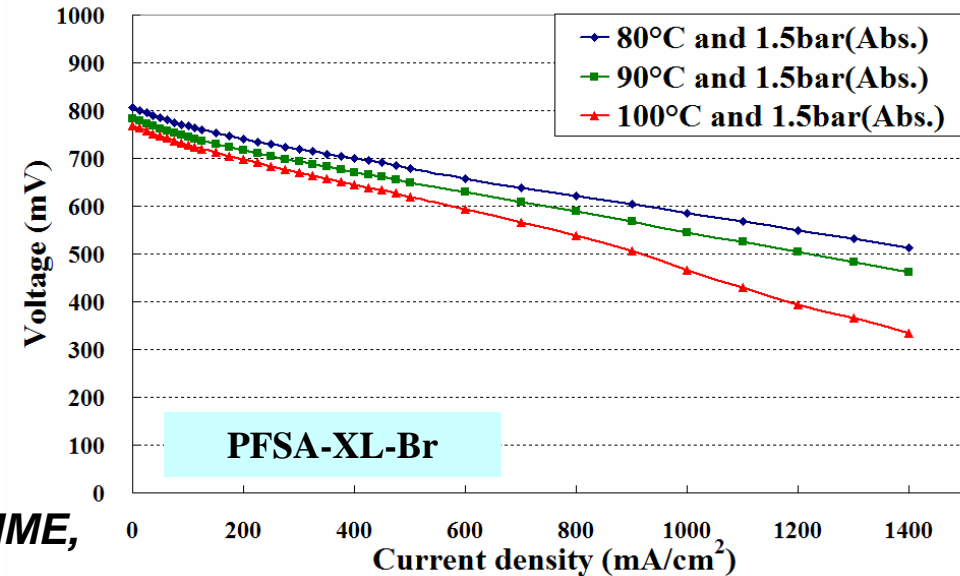
Hirokazu Nakayama, et al. *J. Mater. Chem.*, 1997, 7(6), 1063~1066

Single cell characterization

- Electrodes: E-TEK
- 30% Pt/C; 0.5 mg Pt/cm² ; 0.6 mg Nafion / cm²
- Humidified H₂, O₂
- Controlling temperature and Pressure

H₂/O₂ 1.5 bar (Abs) with hydration (H₂:80 °C - O₂ : 64 °C)

MEA: made by Mr. Marc DUPONT, CNRS-AIME, Montpellier, France



Conclusions

- ◆ *Two cross-linking approaches have been described for PFSA, with macroscopic and local characterisation of properties*
- ◆ *Cross-linking stabilises high IEC/low EW membranes against excessive swelling*
 - *lowers the water uptake and dimensional change*
 - *improves the mechanical properties*
- ◆ *High proton conductivity is maintained in cross-linked and composite membranes*
- ◆ *Preliminary MEA characterisation shows that the presence of the ZrP component allows higher temperature operation, to 120 °C*

Acknowledgements

Project no: 039016

IPHE_GENIE: *International Partnership for a Hydrogen Economy for GENeration of New Ionomer membranEs*

Project partners

- *Energy Research Centre of the Netherlands, Holland*
- *Shanghai Jiao Tong University, China*
- *Dongyue Shenzhou New Materials, China*
- *Fuma –Tech GmbH, Germany*
- *CNRS-University of Montpellier 2, France*
- *Boreskov Institute of Catalysis, Russia*

Thank you
for your attention!