

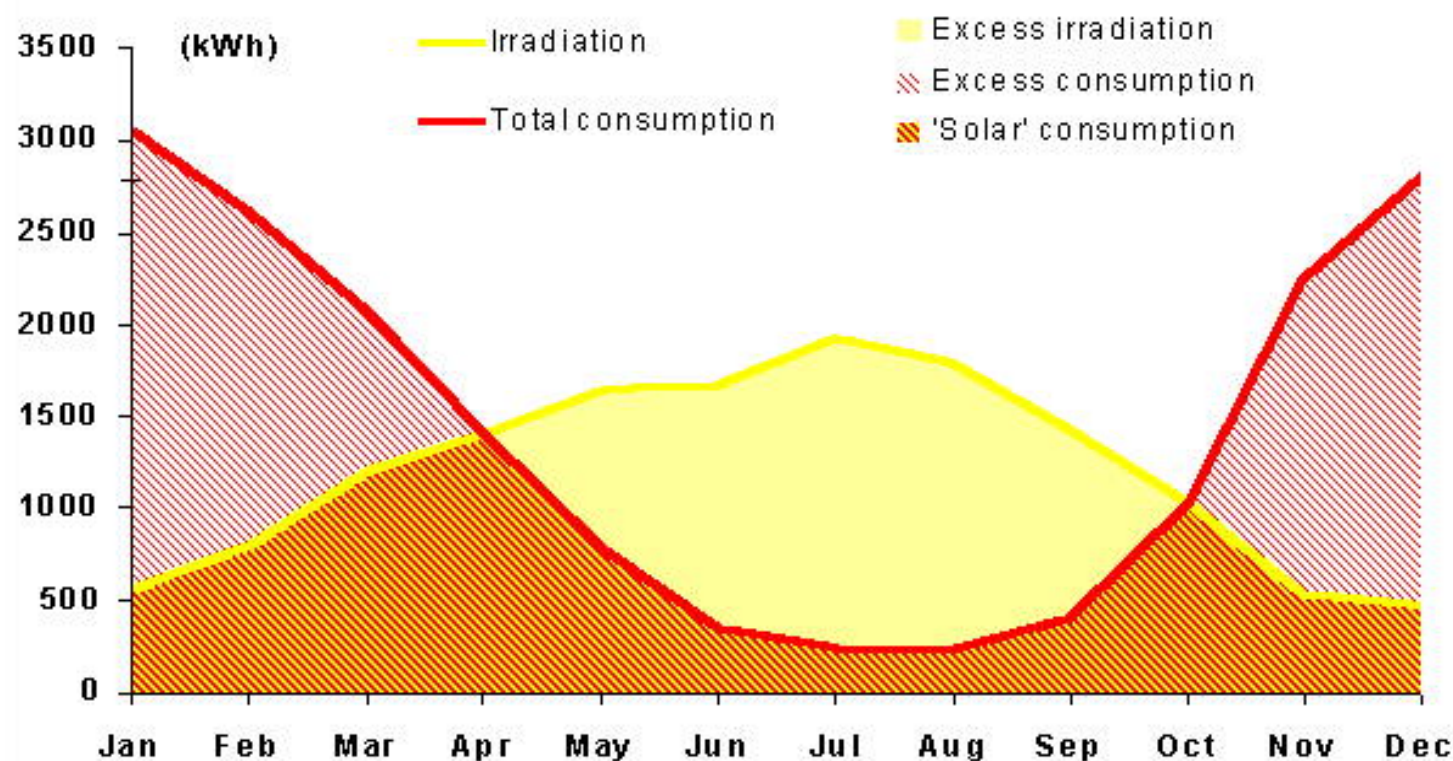
4224

Compact Thermal Energy Storage: Material Development for System Integration

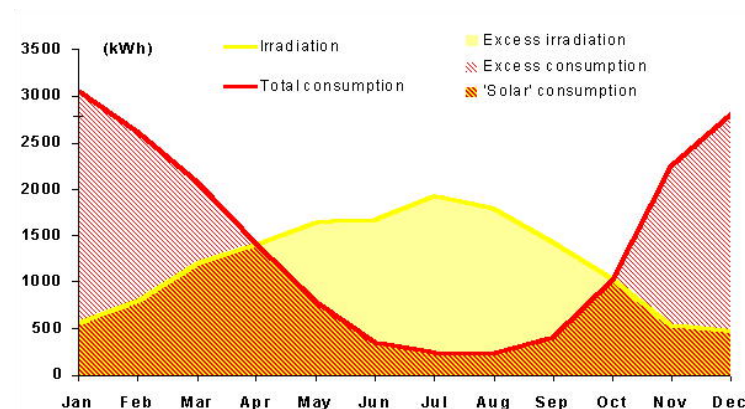
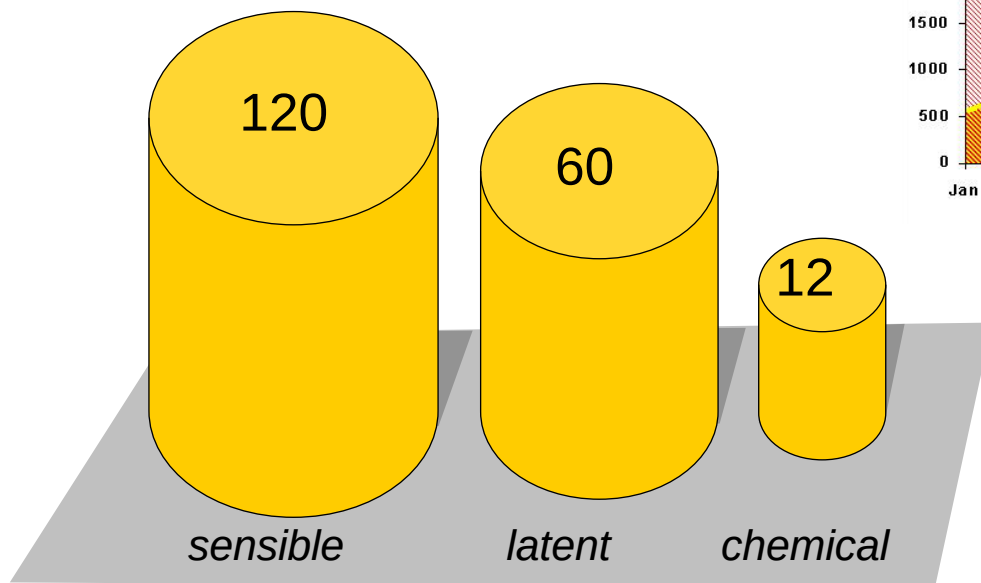
Wim van Helden, Marco Bakker



Seasonal storage of solar heat



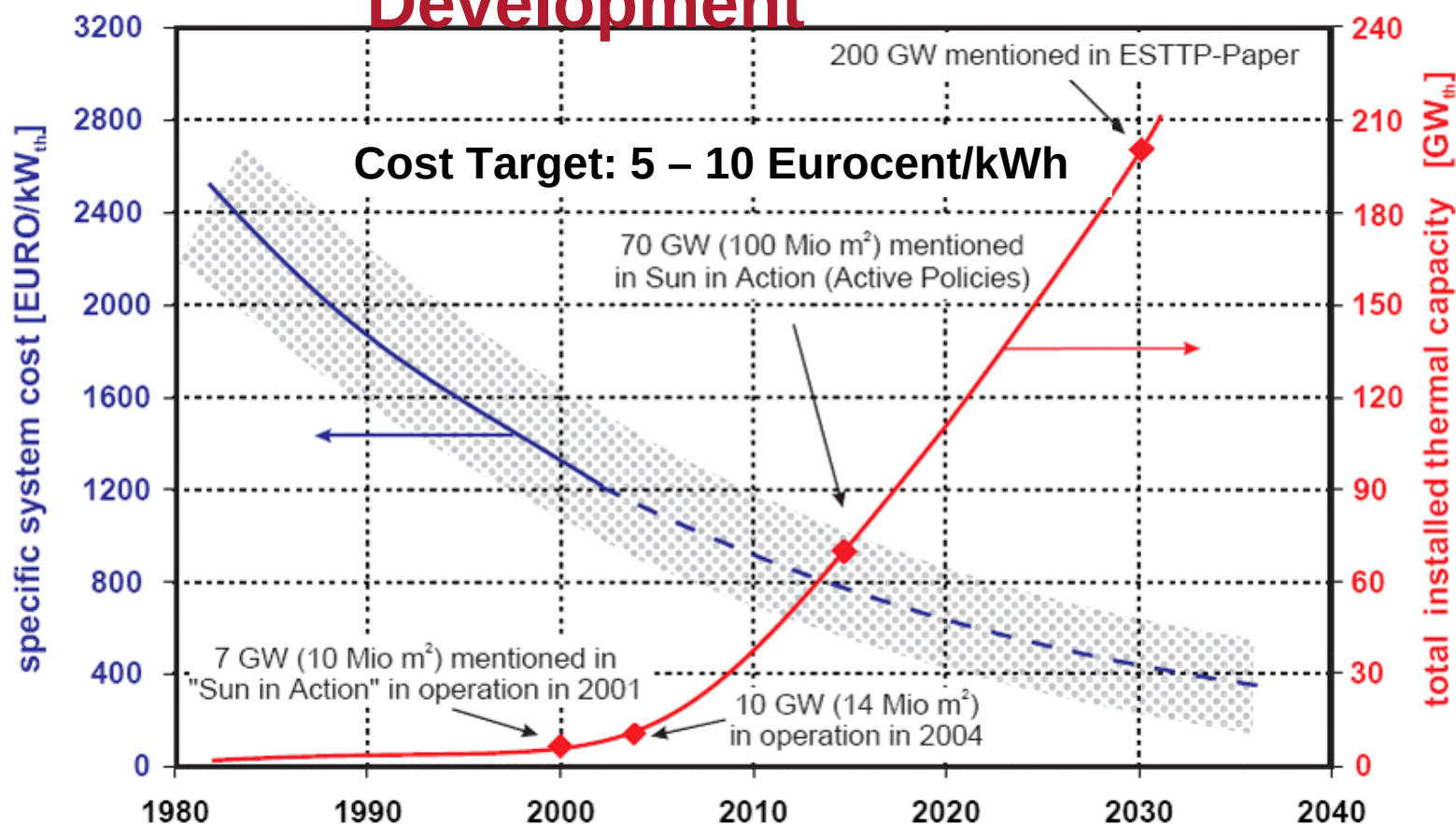
Seasonal storage of solar heat



Seasonal storage volume (m³) needed for a very energy efficient household

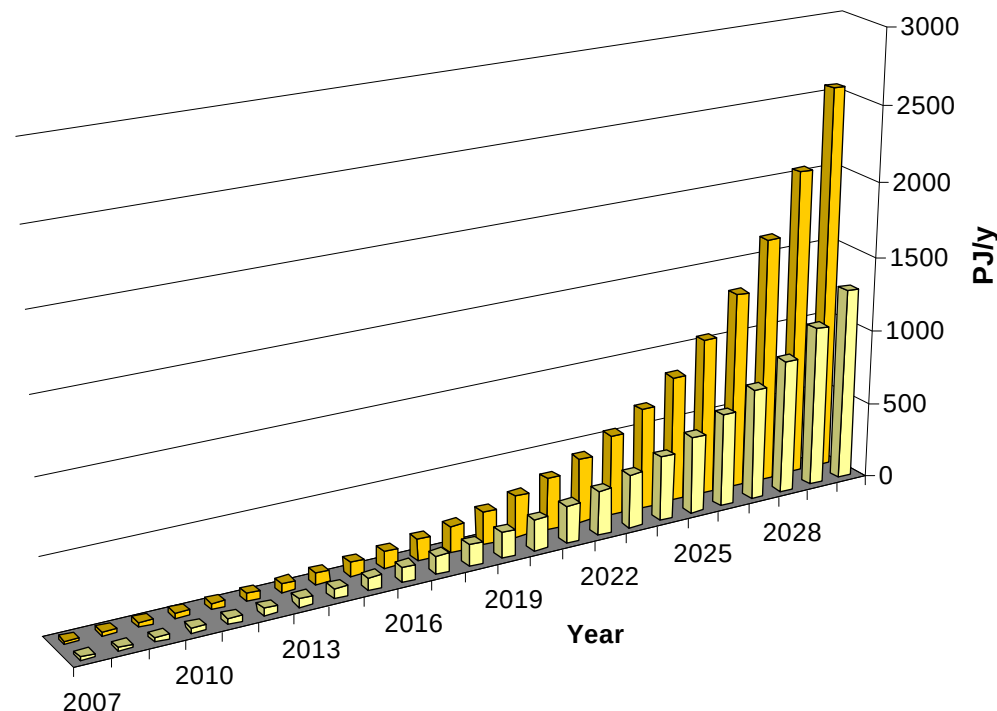


ESTTP Target: Cost Reduction by Economies of Scale and Technological Development

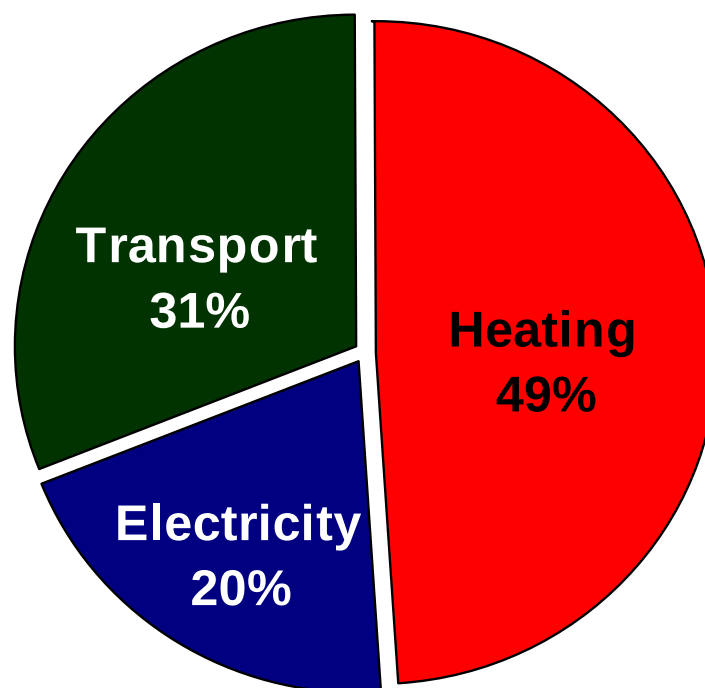


Solar Thermal: Long-term effect of compact heat storage

- Gradual introduction of improved compact storage technologies starting in 2015
- Generated solar thermal energy in EU doubles in 2030



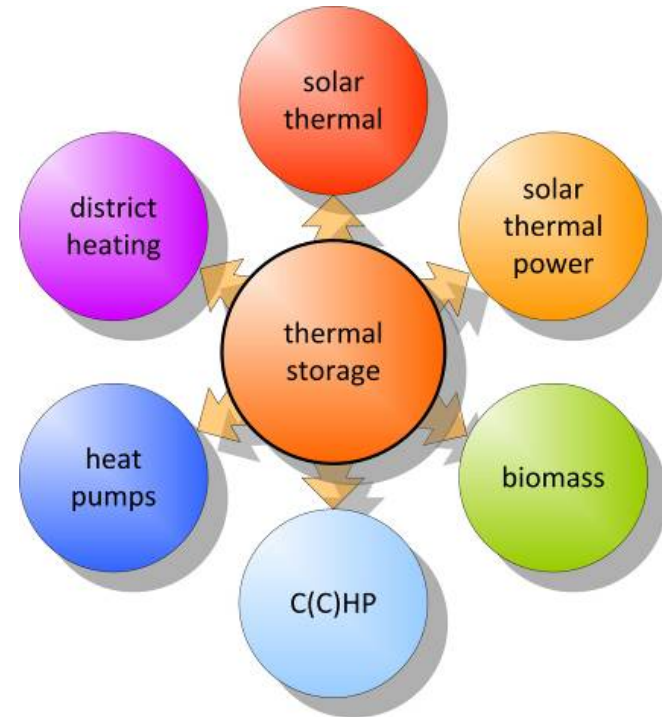
Heating and Cooling: dominant for primary energy consumption



Breakdown of final energy
demand in EU25

Starting position for Thermal Energy Storage

- Thermal storage: enabling technology
 - solar thermal
 - concentrated solar power
 - biomass
 - cogeneration
 - heat pumps
 - district heating
 - ...
- Main bottleneck: materials research
 - new materials needed
 - large gap between materials and applications experts



Materials and Applications

Two International Energy Agency (IEA) programs:

Energy Conservation through
Energy Storage



Solar Heating and Cooling



Task 42/24: Compact Thermal Energy Storage: Material Development for System Integration

Joint Task between Solar Heating and Cooling (SHC) and
Energy Conservation through Energy Storage (ECES)

Operating Agents:

- SHC: Wim van Helden, ECN (NL)
- ECES: Andreas Hauer, ZAE Bayern (DE)

January 2009 – December 2012

Main added value:

Bring together experts from applications *and* material science

Objectives

- Identify, design and develop new materials and composites
- Develop measuring and testing procedures
- Improve performance, stability, and cost-effectiveness
- Develop multi-scale numerical models
- Develop and demonstrate novel storage systems
- Assess the impact of new materials on systems performance
- Disseminate the acquired knowledge and experience
- Create an active and effective research network



Scope

Classes of materials:

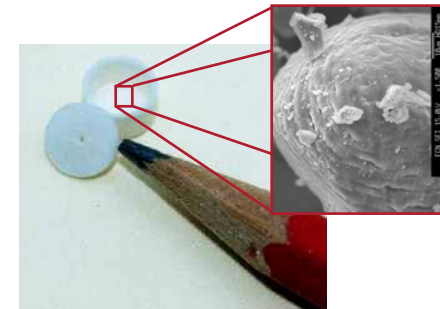
- phase change materials
- sorption and thermochemical materials

From small to large-scale:

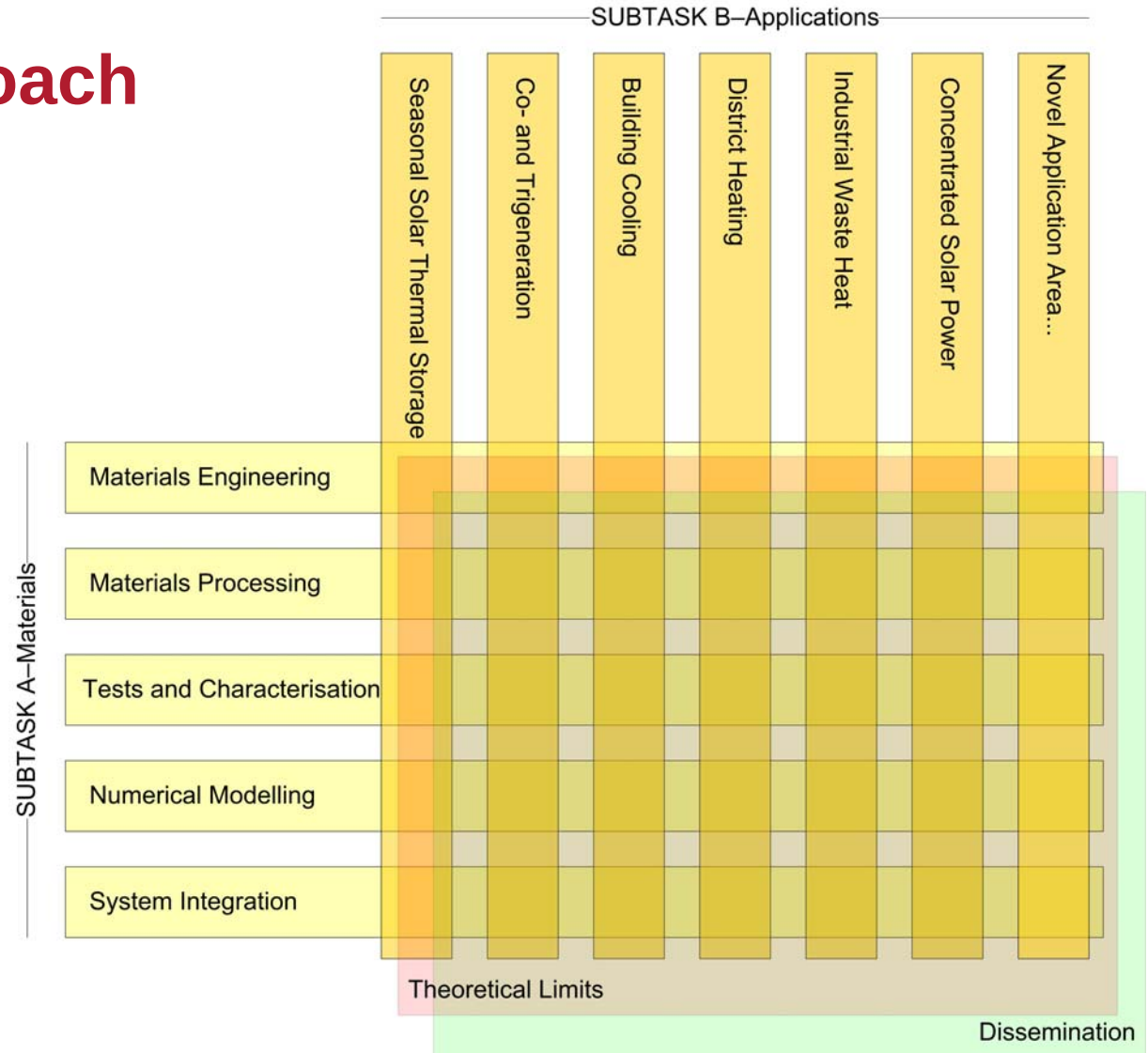
- molecular/particle
material synthesis, micro-scale mass transport, sorption reactions, ...
- bulk
heat/mass transport, wall-material interactions, reactor design, ...
- system

economical feasibility studies, case

studies, system tests, ...



Matrix approach

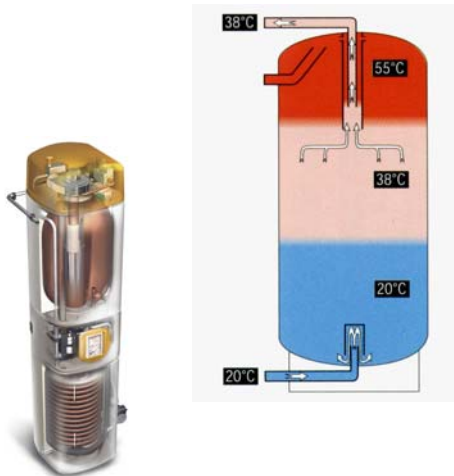


Participation and work

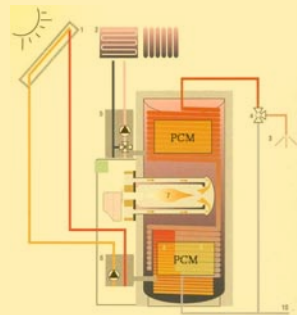
- 9 Countries that have funded participation:
 - AT (2 yr), AU, CH (1 yr), DE, ES, FR, NL, SE, UK
- 16 Organisations
- This number will grow after the start 1 January 2009
- International and national project brought into the Task
- Until now > 850 person months in projects
- Expert meeting + workshop twice per year
- Kick-off: 11-13 February 2009, Bad Tölz, Germany

Stage of development of Thermal Energy Storage technologies

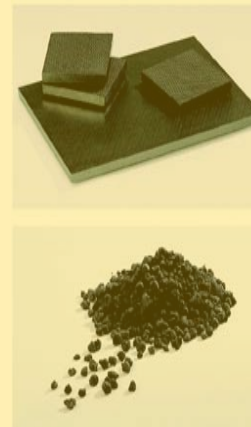
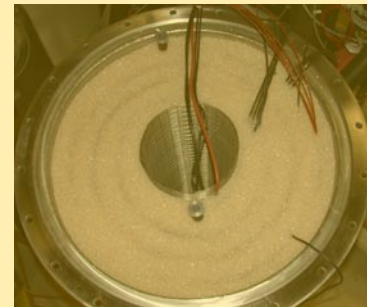
Water (sensible)
Mature market



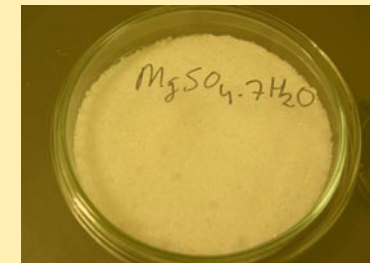
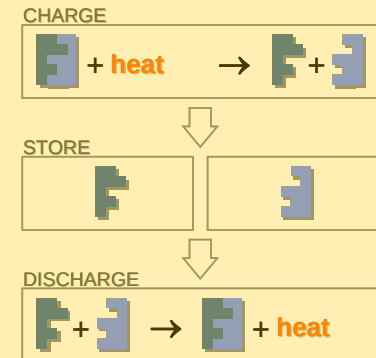
PCM (latent)
Demonstration



Sorption (physical)
Development



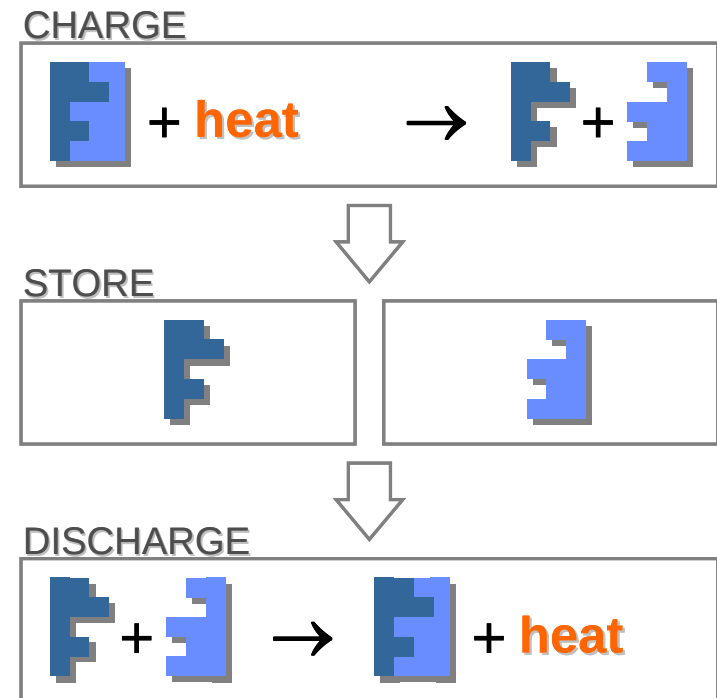
TCM (chemical)
Research



Chemical heat storage

General principle $A + B \rightleftharpoons AB + \text{heat}$

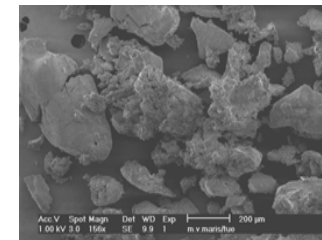
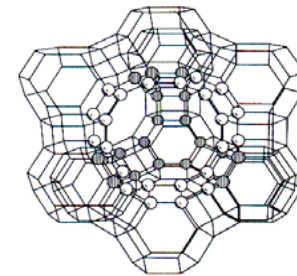
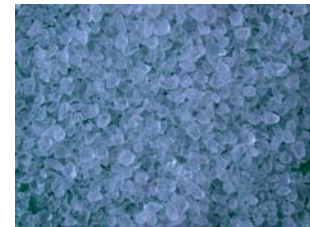
- Components stored separately without heat loss
- Long term heat storage
- Charge with temperatures typically higher than 100°C
- Storage capacity between 250 and 4000 kJ/kg



New Materials and Composites

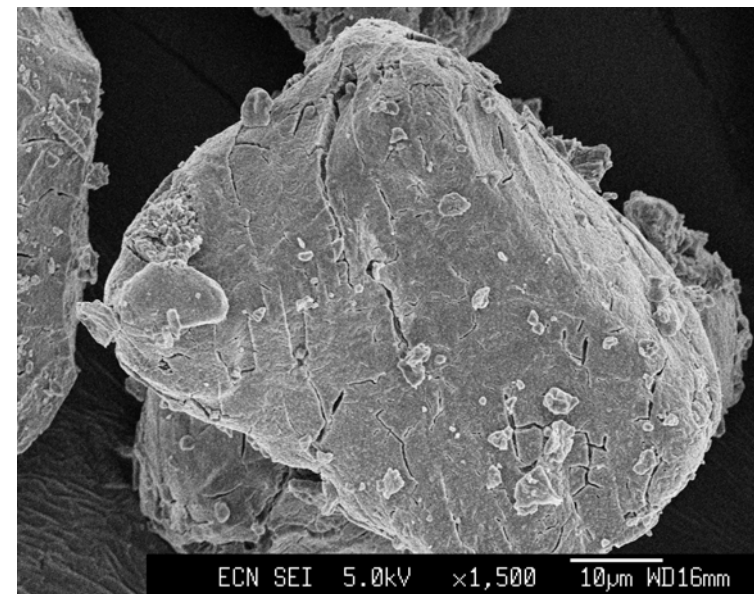
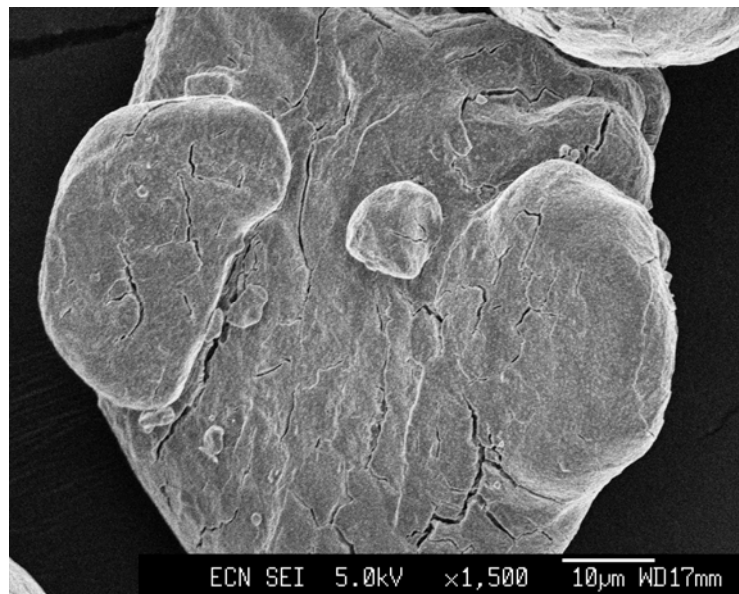
Phase-change, sorption, thermochemical

- Improve active materials (silicagels, zeolites, hydrates)
- Novel materials (phase-change, thermochemical)
- Active material with carrier material (zeolites on aluminum)
- Combination of active materials (zeolites plus hydrates)

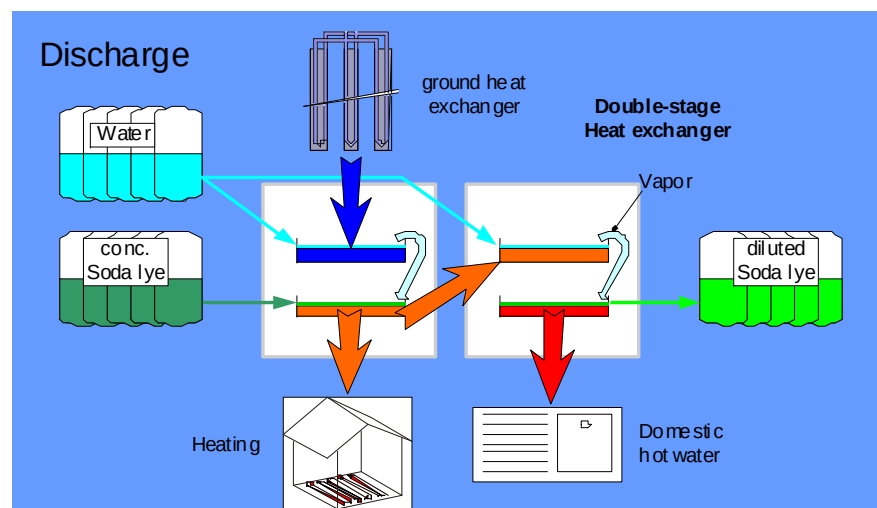
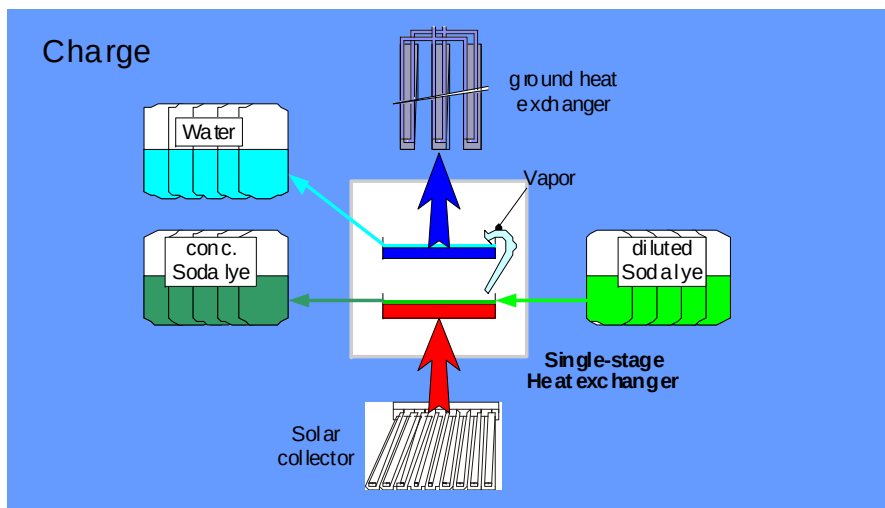


Morphology

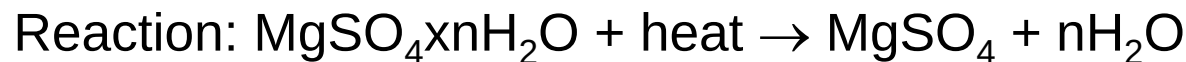
Marked difference found in hydration behaviour of samples of two different suppliers



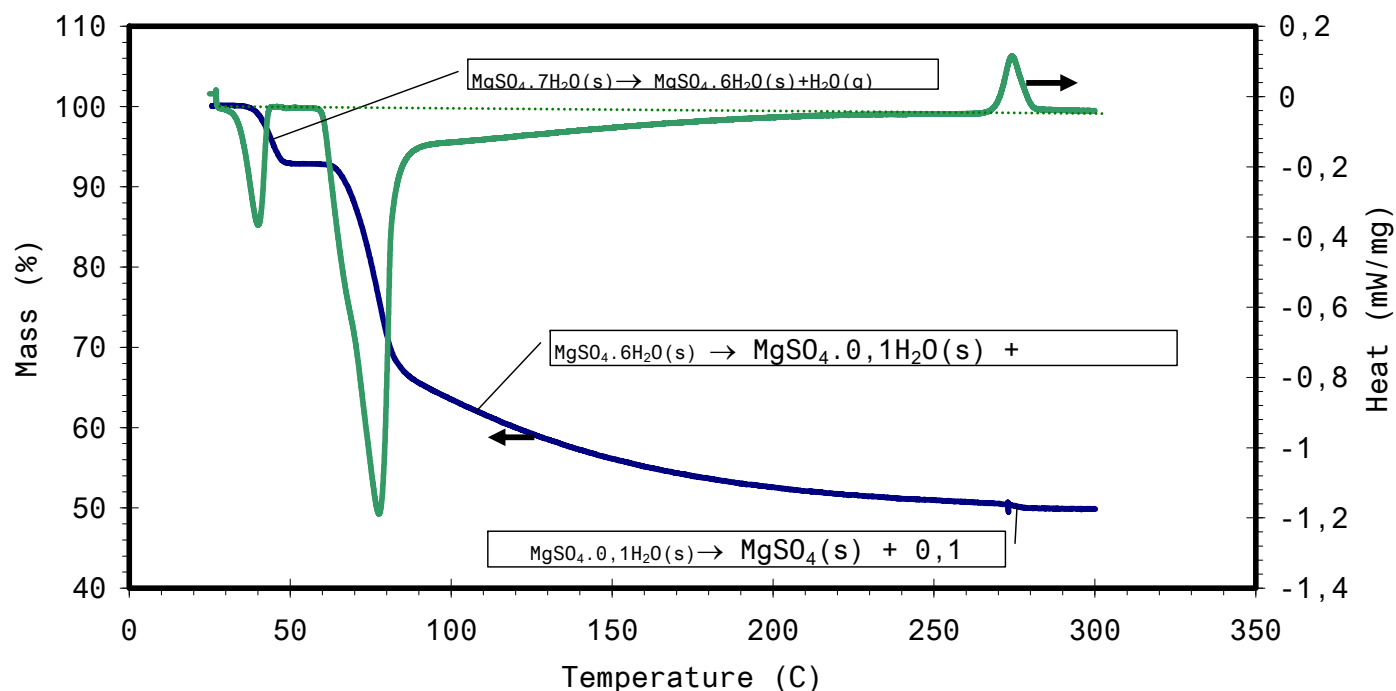
Sodiumhydroxide Storage (EMPA – CH)



Thermochemical material – $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$



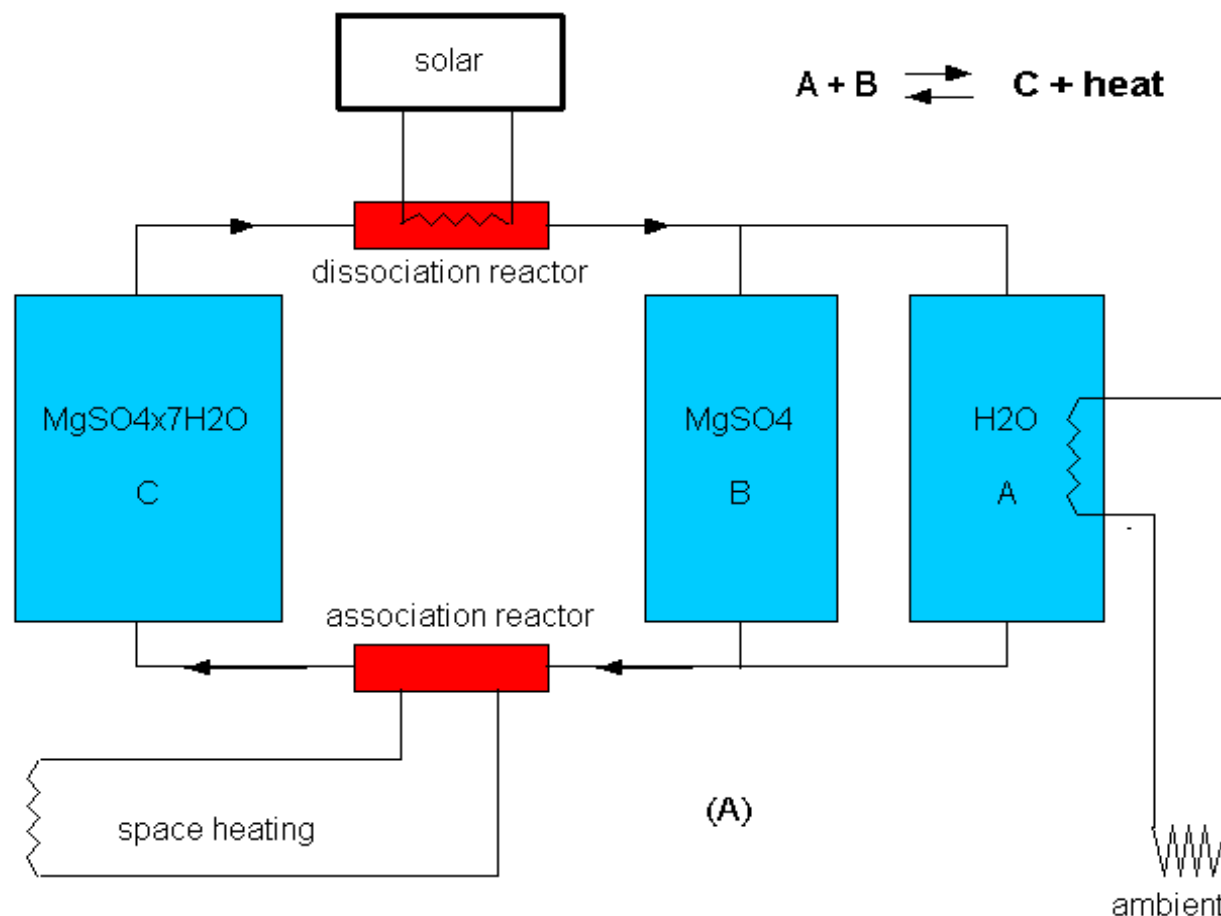
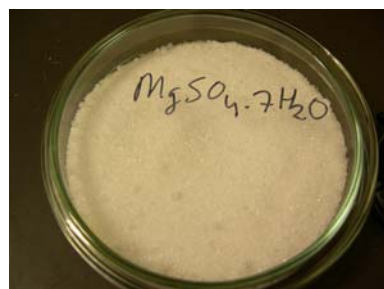
Dehydration of $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$
(10 mg, 1 C/min, $\text{N}_2 + \text{H}_2\text{O}$ atmosphere with RH=40%)



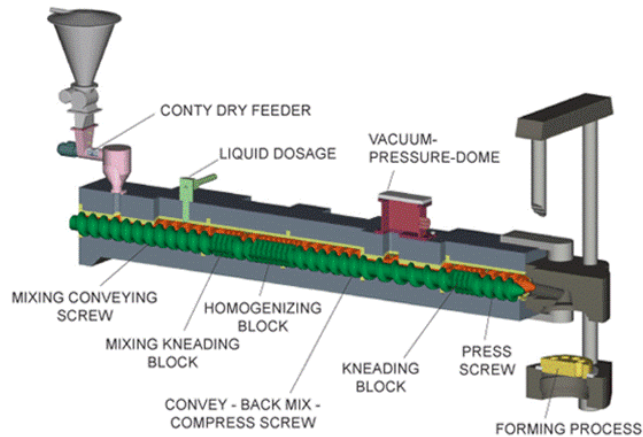
Sample mass decreases at increasing temperature

Magnesiumsulphate (ECN –NL)

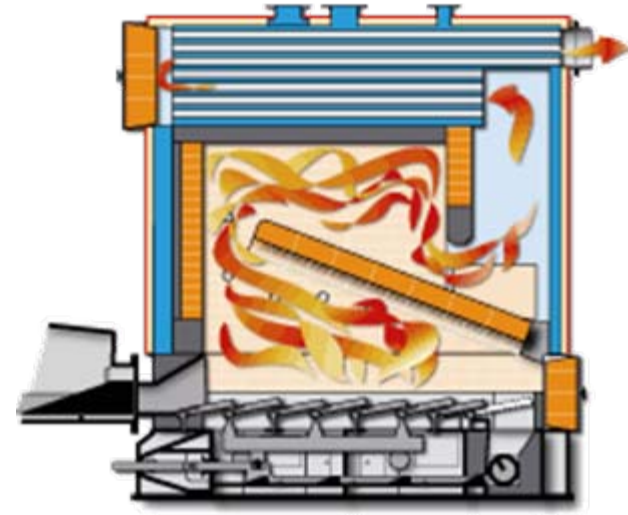
Separate
reactor
concept



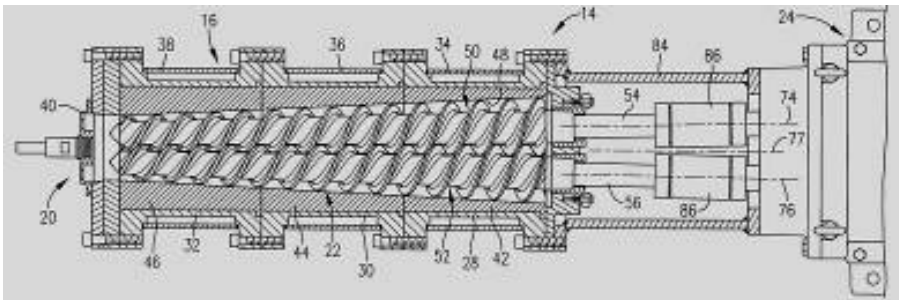
Reactor concepts



Hot screw dryer, extruder



Moving grate dryer



What You Can Do For Compact TES

- Do materials research
 - Do applications research
 - Find connected fields
 - Incorporate it into your strategy
 - Convince policy makers
 - Support a programmed international effort
 - Spread the word
-
- Contact us: www.iea-shc.org/task42

The First Thing You Can Do

Memorise the number